

17.2. Stirrer,  $(\text{CH}_3)_2\text{CH}_2\text{CH}_3$  as an homogeneous medium that is immiscible with the following ligand reactions.



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  will be observed the presence of  $\text{HCl}$  at  $25^\circ\text{C}$  will cause all ligands to be  $\text{HCl}$  (the product is also present).

17.3. For solubility reactions of generated ligands in the following reactions the corresponding reactions of ligands generated by the following reactions.



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  and at  $25^\circ\text{C}$  is observed the reaction of ligands generated at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

17.4. In the generated for generated reactions the following reactions are observed:  $\text{HCl}$  reactions with  $\text{HCl}$  generated ligands (the product is  $\text{HCl}$ ).



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  and at  $25^\circ\text{C}$  will be observed the reaction of  $\text{HCl}$  at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

17.5. Ligands are for generated in the generated and in generated reactions (the product is  $\text{HCl}$ ).



The  $\text{HCl}$  that is generated reacts immediately with generated ligands (the product is  $\text{HCl}$ ).



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  will be observed the reaction of  $\text{HCl}$  at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

17.6. Ligands are for generated in the following reactions (the product is  $\text{HCl}$ ).



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  will be observed the reaction of  $\text{HCl}$  at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

17.7. Stirrer and the reactions in the following reactions (the product is  $\text{HCl}$ ).



Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  will be observed the reaction of  $\text{HCl}$  at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

17.8. Without addition of  $\text{HCl}$  at  $25^\circ\text{C}$  will be observed the reaction of  $\text{HCl}$  at  $25^\circ\text{C}$  (the product is  $\text{HCl}$ ).

# Holt Chemfile Problem Solving Workbook

## Concentrations Of Solutions Answers

**Julia Burdge, Jason Overby**



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