

HISTORY (027)

Class XII, Syllabus (2014-15)

Paper One

100 Marks

Units	Periods	Marks
Themes in Indian History Part-I Units 1 - 4	55	25
Themes in Indian History Part-II Units 5 - 9	65	25
Themes in Indian History Part-III Units 10 - 15	80	25
Unit 16 : Map Work	10	5
Project work	10	20
	220	100

Note: There is no change in the syllabus. Value Based Question can be from Part-I, 2, 3 textbooks and carry 04 marks. Accordingly teacher can reduce weightage of the corresponding section.

Class XII: Themes in Indian History	
Themes	Objectives
PART - I (Periods 55) 1. The Story of the First Cities: Harappan Archaeology. (13) Broad overview: Early urban centres. Story of discovery: Harappan civilisation Excerpt: Archaeological report on a major site. Discussion: How it has been utilized by archaeologists/historians. 2. Political and Economic History: How Inscriptions tell a story. (14) Broad overview: Political and economic history from the Mauryan to the Gupta period. Story of discovery: Inscriptions and the decipherment of the script. Shifts in the understanding of political and economic history. Excerpt: Asokan inscription and Gupta period land grant. Discussion: Interpretation of inscriptions by historians.	➤ Familiarize the learner with early urban centres as economic and social institutions. ➤ Introduce the ways in which new data can lead to a revision of existing notions of history. ➤ Illustrate how archaeological reports are analyzed and interpreted by scholars. ➤ Familiarize the learner with major trends in the political and economic history of the subcontinent. ➤ Introduce inscriptional analysis and the ways in which these have shaped the understanding of political and economic processes.

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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect to zero energy at zero displacement from the spring s unstretched length $x = 0$ is $u = \frac{1}{2} kx^2$ figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the Δh 2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δh is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web pertanyaan serupa iklan perhatikan diagram energi potensial berikut pada diagram di atas energi aktivasi untuk reaksi baliknya ditandai dengan

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web answer see below explanation the rate of a chemical reaction a b c d is given in first approximation by $v = k e^{-E_a/RT}$ where k depends by many factors as solvent geometry type of reaction E_a is the activation energy R is the universal constant of gases T is the absolute temperature

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web mar 17 2016 3 what is the activation energy of a reaction and how is this energy related to the activated complex of the reaction E_a is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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web key points potential energy is energy that has the potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate Δh for the reaction and label it with the correct value see attached image

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web aug 16 2019 1 answer sorted by 4 the activation energy is not the average kinetic energy that the colliding particles need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label pe diagrams for exothermic and endothermic reactions and what effect adding a catalyst or inhibitor has on the diagram

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web may 1 2013 a potential energy diagram shows the change in potential energy of a system as reactants are converted into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change Δh is positive for an endothermic reaction and negative for an

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web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or endothermic 2 what is the activation energy for the forward reaction 3 what is the