Atomic History Worksheet Answer Key

Democritus Proposes the Atom

- 1. The word atom comes from a Greek word that means indivisible
- 2. Which of the following statements is part of Democritus's theory about atoms?
 - Atoms are the smallest piece of matter.
 - d. Atoms exist but will never by fully proven.
- 3. We know that Democritus was right about atoms. So why did people ignore Democritus's ideas for such a long time?

Democritus's ideas were ignored for a long time because he had no proof of the atoms existence and people believed Aristotle's ideas.

Dalton Creates an Atomic Theory Based on Experiments

- 4. By conducting experiments and making observations, Dalton figured out that elements combine in random proportions because they're made of individual atoms. False
- 5. Dalton's theory states that atoms cannot be changed, created, or destroyed.
- 6. Atoms of different elements are exactly alike. False
- 7. How did Dalton think atoms formed new substances?

Atoms can be combined in set proportions to form new substances.

Thomson Finds Electrons in the Atom

- Mark the following statements True or False.
 - In 1897, J. J. Thomson made a discovery that proved Dalton's atomic model was correct. False
 - 9. Thomson discovered that there were small particles inside the atom. True
 - 10. Thomson found that the electrically charged plates affected the direction of a cathode-ray tube beam. True
 - 11. Thomson knew the beam was made of particles with a positive charge because it was pulled toward a positive charge. False
 - 12. The two types of charges are positive and neutral. False

History Of The Atom Answer Key

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