- What are moleculed Give examples.
- What is a physical change and littlesamples of physical changes.
- What is a chemical change and give examples of chemical changes.

2.2 Main Body (Leason per sentation)

Educator starts lesson off with an exciting demonstration as mentioned in the introduction and explains the relevant concepts in terms of a chemical change. [30 min.]

- Matter it all around us and it undergoes changes all the time. Mese changes can be classified as PHYSICAL CHANGES OR CHENECAL CHANGES.
- The CONCEPTS of PHYSICAL CHANGE and CHEMICAL CHANGE is shown in the table below, educator engages learners in a discussion of the different aspects of physical and chemical change as mentioned in the table to re-enforce the concepts.

physical changes	e beminal e hanges
common signs that a chemical change has occurred are:	common signs that a physical change has occurred are:
I. Production of gas bubbles C. Change in the way something smelt I. A release of energy such as a flash or a sound (like a firecracker) I. A precipitate forms (two liquids mixed together form a solid and a liquid)	I. change in the size I. change in shape. I. change in colour, or I. Change in state/ phase of matter of a substance. I. No new substance is produced.
humples of physical changes	Facompile a of e he mic al e hange a
A grape when Hepped on (changes shape) Blowing up a balloon (changes size and shape) Liquid water furning to ice (changes state of matter) Liquid water furning to steam (changes state of matter) Missing salt and sugar (changes the appearance, but you can still reparate the mixture) Missing water and salt (changes the appearance, but you can still reparate the mixture)	Metal sisting prew substance formest; Stamach algesting food (break down of food to new substances) Plant's anying out photosynthesis (putting water and carbon dioxide together to make sugar) Mixing boiling sodo and vinegar (makes a neutral liquid and a gas)

- Educator documes the results of the experiment with the learners.
- The liquid hydrogen peroxide decomposes (breaks up) to hirm oxygen gat and liquid water (the atomic model the using ball and stock as well as space filling diagrams to show the reactants and products in this reaction).
- The MANGANESE DIGKDE is a catalyst and speeds up the reaction. A CARDING a chemical substance that changes the speed of a reaction without getting used up in the
 process, of the end of the reaction the catalyst can be recovered. Eg. ENCYMES in our body help to speed up the digestive process.

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Wolfgang Guggemos

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- 1. Understanding the eBook Grade 10 Caps Physical Sciences Lesson Plan
 - The Rise of Digital Reading Grade 10 Caps Physical Sciences Lesson Plan
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 10 Caps Physical Sciences Lesson Plan
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an Grade 10 Caps Physical Sciences Lesson Plan
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 10 Caps Physical Sciences Lesson Plan
 - Personalized Recommendations
 - o Grade 10 Caps Physical Sciences Lesson Plan User Reviews and Ratings
 - Grade 10 Caps Physical Sciences Lesson Plan and Bestseller Lists
- 5. Accessing Grade 10 Caps Physical Sciences Lesson Plan Free and Paid eBooks
 - Grade 10 Caps Physical Sciences Lesson Plan Public Domain eBooks
 - o Grade 10 Caps Physical Sciences Lesson Plan eBook Subscription Services
 - Grade 10 Caps Physical Sciences Lesson Plan Budget-Friendly Options
- 6. Navigating Grade 10 Caps Physical Sciences Lesson Plan eBook Formats

- ePub, PDF, MOBI, and More
- Grade 10 Caps Physical Sciences Lesson Plan Compatibility with Devices
- Grade 10 Caps Physical Sciences Lesson Plan Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Grade 10 Caps Physical Sciences Lesson Plan
 - Highlighting and Note-Taking Grade 10 Caps Physical Sciences Lesson Plan
 - Interactive Elements Grade 10 Caps Physical Sciences Lesson Plan
- 8. Staying Engaged with Grade 10 Caps Physical Sciences Lesson Plan
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Caps Physical Sciences Lesson Plan
- 9. Balancing eBooks and Physical Books Grade 10 Caps Physical Sciences Lesson Plan
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Grade 10 Caps Physical Sciences Lesson Plan
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 10 Caps Physical Sciences Lesson Plan
 - Setting Reading Goals Grade 10 Caps Physical Sciences Lesson Plan
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 10 Caps Physical Sciences Lesson Plan
 - Fact-Checking eBook Content of Grade 10 Caps Physical Sciences Lesson Plan
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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