

Name: _____ Date: _____

pH and pOH

Choose the correct options:

What is the pH of a neutral solution at 25°C?

- (a) 0
- (b) 7
- (c) 14
- (d) 3

Which of the following substances has a high pOH value (i.e., more basic)?

- (a) Lemon juice (pH 2)
- (b) Baking soda solution (pH 9)
- (c) Milk (pH 6)
- (d) Water (pH 7)

The pH of a solution is 4. What is the approximate concentration of hydrogen ions $[H^+]$ in the solution?

- (a) $1 \times 10^{-4} M$
- (b) $1 \times 10^{-10} M$
- (c) $1 \times 10^{-1} M$
- (d) $1 M$

Which of the following statements is true about the pOH scale?

- (a) The pOH of a neutral solution is 0.
- (b) The pOH scale ranges from 0 to 14, just like the pH scale.
- (c) A pOH of 7 is acidic.
- (d) pOH can never be negative.

A solution has a pOH of 3. What is the pH of this solution?

- (a) 3
- (b) 7
- (c) 11
- (d) 14

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