

9 • Bonding & Molecular Structure

PRACTICE TEST

1. The correct Lewis symbol for ground state carbon



2. The correct Lewis symbol for ground state aluminum is



3. Using the picture below, what process corresponds to the lattice energy?



4. Which of the following favors formation of an ionic compound?

- (a) low ionization energy for metal
(b) high electron affinity for non metal
(c) high lattice energy
(d) all of a-c above
e) none of a-c above

5. Which of the atoms below is least likely to violate the octet rule?

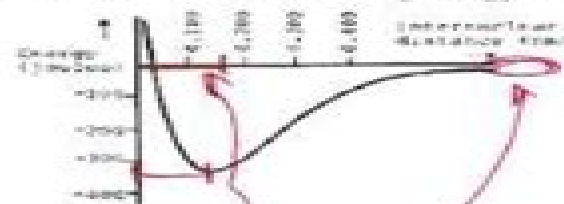
- a) Be b) P c) S d) B (e) F

6. How many electrons are shown in the Lewis structure of perchlorate ion, ClO_4^- ?

- a) 30 b) 31 (c) 32 d) 50 e) 51



- Questions 7 - 9 refer to the following energy diagram:



7. What is the energy of the two isolated atoms?

- a) -400 J b) -335 J (c) 0 J d) 0.100 nm e) -155 J

8. What is the bond length of the bond between the two atoms?

- a) 0.020 nm b) 0.140 nm (b) c) 0.400 nm d) 0.330 nm e) 2.00 nm

9. What is the bond energy (bond strength) of the bond?

- a) 0 J b) -110 J (d) -330 J c) -10 J e) -422 J

Should this be (+) value? Minus --

10. What is the bond order of the C-O bond in acetone?



- a) 4 b) 1.5 c) 0.5 d) 1 (c) 2

double = B.O 2

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1. Understanding the eBook Mcdougal Geometry Chapter2 Test Answer Key
 - The Rise of Digital Reading Mcdougal Geometry Chapter2 Test Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcdougal Geometry Chapter2 Test Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcdougal Geometry Chapter2 Test Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcdougal Geometry Chapter2 Test Answer Key
 - Personalized Recommendations

- Mcdougal Geometry Chapter2 Test Answer Key User Reviews and Ratings
- Mcdougal Geometry Chapter2 Test Answer Key and Bestseller Lists
- 5. Accessing Mcdougal Geometry Chapter2 Test Answer Key Free and Paid eBooks
 - Mcdougal Geometry Chapter2 Test Answer Key Public Domain eBooks
 - Mcdougal Geometry Chapter2 Test Answer Key eBook Subscription Services
 - Mcdougal Geometry Chapter2 Test Answer Key Budget-Friendly Options
- 6. Navigating Mcdougal Geometry Chapter2 Test Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcdougal Geometry Chapter2 Test Answer Key Compatibility with Devices
 - Mcdougal Geometry Chapter2 Test Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcdougal Geometry Chapter2 Test Answer Key
 - Highlighting and Note-Taking Mcdougal Geometry Chapter2 Test Answer Key
 - Interactive Elements Mcdougal Geometry Chapter2 Test Answer Key
- 8. Staying Engaged with Mcdougal Geometry Chapter2 Test Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcdougal Geometry Chapter2 Test Answer Key
- 9. Balancing eBooks and Physical Books Mcdougal Geometry Chapter2 Test Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcdougal Geometry Chapter2 Test Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mcdougal Geometry Chapter2 Test Answer Key
 - Setting Reading Goals Mcdougal Geometry Chapter2 Test Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcdougal Geometry Chapter2 Test Answer Key
 - Fact-Checking eBook Content of Mcdougal Geometry Chapter2 Test Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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