

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

### Skills Worksheet

## Concept Review

### Section: How are Elements Organized?

Answer the following questions in the space provided.

1. Why do Li, Na, K, Rb, Cs, and Fr all react with Cl in a 1:1 ratio forming substances with similar properties?

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2. Explain the method that John Newlands used to organize the elements.

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3. What method did Dmitri Mendeleev use to arrange his periodic table?

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4. Why did Mendeleev have gaps in his table? How did he use these gaps?

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5. What was Henry Moseley's contribution to the periodic table?

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### Concept Review continued

6. Why was Moseley able to resolve the discrepancy Mendeleev could not?

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7. Explain the importance of valence electrons.

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8. Why do elements with similar properties appear in a periodic table?

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9. How is the electron configuration similar for each group?

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10. How is the electron configuration similar for each period?

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# Holt Chemistry Concept Review Answers Chapter 6

**Kaplan Test Prep**



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
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