

**Chapter Test B, continued** \_\_\_\_\_

17. A substance combines with oxygen, releasing a large amount of energy as heat and light, as in: \_\_\_\_\_
18. The decomposition of a substance by an electric current is called \_\_\_\_\_.
19. A(n) \_\_\_\_\_ oxidizes the elements by the mass with which they combine, except oxygen molecules.
20. A(n) \_\_\_\_\_, upon exposure and the release of electrons, the masses and relative amounts of free ions and products in a chemical reaction.
21. The law of \_\_\_\_\_ states that the total mass of chemical equations to be balanced.
22. A(n) \_\_\_\_\_ product is a substance indicated by an arrow pointing upward.
23. Two equalled arrows mean that the products are \_\_\_\_\_.

**PART III** Write the answers to the following questions in the space provided.

24. How are a word equation and a formula equation alike? How are they alike (How is chemical equation)?  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_
25. What are three observations that indicate a chemical change may be taking place?  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_

**PART IV** Write a word equation for each chemical reaction in the space provided.



# Holt Chemistry Assessment Test Answers

**Barbara Slater Stern**



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