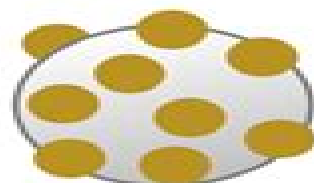


Dalton, 1808

First to describe atoms in a modern, scientific sense

– Doesn't explain electricity

+ Idea of "atoms"

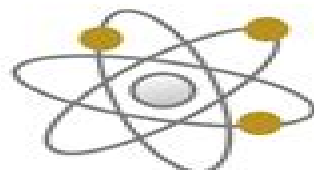


Thomson, 1897

Thomson's Plum Pudding Model

– Doesn't explain why some of Rutherford's α -particles bounced back

+ Protons & electrons



Rutherford, 1911

Rutherford shot α -particles through gold foil; some bounced back!

– Why don't the electrons lose energy and crash into the nucleus?

+ the Nucleus

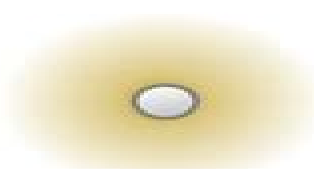


Bohr, 1913

Basis for our modern atomic model

– Doesn't explain quantum mechanics

+ Electron Shells



Schrödinger, 1926

Quantum mechanics

– Why are some atoms of the same element heavier?

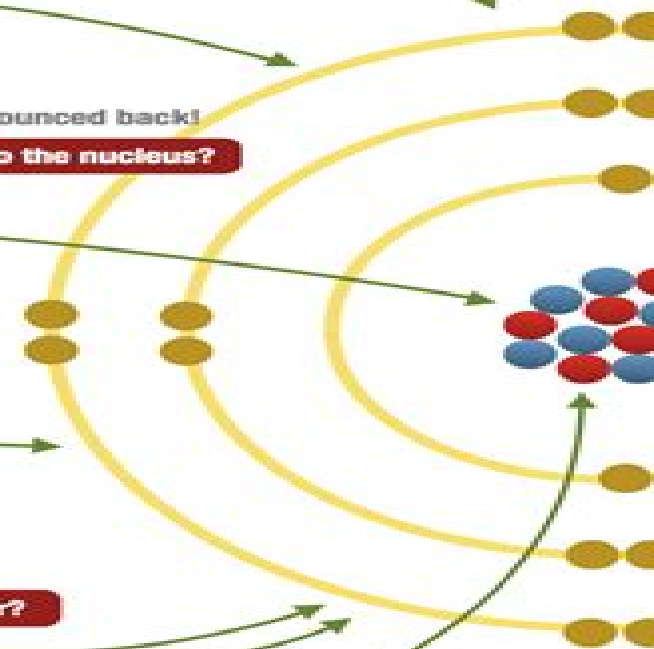
+ Subshells

+ 'Shells' are actually 'orbitals'



Chadwick, 1932

+ Neutrons!



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JA Banks



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Table of Contents History Of The Atom Notes 2012 Tesccc

1. Understanding the eBook History Of The Atom Notes 2012 Tesccc
 - The Rise of Digital Reading History Of The Atom Notes 2012 Tesccc
 - Advantages of eBooks Over Traditional Books
2. Identifying History Of The Atom Notes 2012 Tesccc
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an History Of The Atom Notes 2012 Tesccc
 - User-Friendly Interface
4. Exploring eBook Recommendations from History Of The Atom Notes 2012 Tesccc
 - Personalized Recommendations
 - History Of The Atom Notes 2012 Tesccc User Reviews and Ratings
 - History Of The Atom Notes 2012 Tesccc and Bestseller Lists

5. Accessing History Of The Atom Notes 2012 Tesccc Free and Paid eBooks
 - History Of The Atom Notes 2012 Tesccc Public Domain eBooks
 - History Of The Atom Notes 2012 Tesccc eBook Subscription Services
 - History Of The Atom Notes 2012 Tesccc Budget-Friendly Options
6. Navigating History Of The Atom Notes 2012 Tesccc eBook Formats
 - ePub, PDF, MOBI, and More
 - History Of The Atom Notes 2012 Tesccc Compatibility with Devices
 - History Of The Atom Notes 2012 Tesccc Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of History Of The Atom Notes 2012 Tesccc
 - Highlighting and Note-Taking History Of The Atom Notes 2012 Tesccc
 - Interactive Elements History Of The Atom Notes 2012 Tesccc
8. Staying Engaged with History Of The Atom Notes 2012 Tesccc
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers History Of The Atom Notes 2012 Tesccc
9. Balancing eBooks and Physical Books History Of The Atom Notes 2012 Tesccc
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection History Of The Atom Notes 2012 Tesccc
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine History Of The Atom Notes 2012 Tesccc
 - Setting Reading Goals History Of The Atom Notes 2012 Tesccc
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of History Of The Atom Notes 2012 Tesccc
 - Fact-Checking eBook Content of History Of The Atom Notes 2012 Tesccc
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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