CHAPTER 8

CHEMICAL EQUILIBRIUM MCQs

Q.1	A reaction is reversible because				
	(a)	reactants are reactive	(b)	products are reactive	
	(c)	products are stable	(d)	reactants are stable	
Q.2	A large value of Kc means that at equilibrium				
	(a)	(a) less reactants and more products			
	(b)	more reactants and less product			
	(c)	same amount	173		
	(d)	none	. 1		
Q.3	Extent to H2 + I2 → 2HI can be increased by				
	(a)	increasing pressure	(b)	increasing product	
	(c)	increasing temp.	Marian.	(d) adding a catalyst	
Q.4	Strength of an acid can be determined by				
	(a)	PKa	(b) PKp	
	(c)	POH	(d)	PKw	
Q.5	In an	exothermic reversible re	eaction i	increase in temp shifts the	
equili	ibrium t	to ot			
	(a)	reactant side	(b)	product side	
	(c)	remains unchanged	(d)	none	
Q.6	Units of Kw are				
	(a)	mole dm-3	(b)	mole2 dm-3	
	(c)	mole2 dm-6	(d)	mole2 dm-3	
Q.7	A basic Buffer solution can be prepared by mixing				
	(a)				
	(b)	strong acid and its salt with weak base			
	(c)	weak base and its salt with strong acid			
	(d)				
Q.8	Buffer action can be explained by				
	(a)	common ion effect	(b)	law of mass action	

Mcq On Chemical Equilibrium

Thomas Griffiths



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