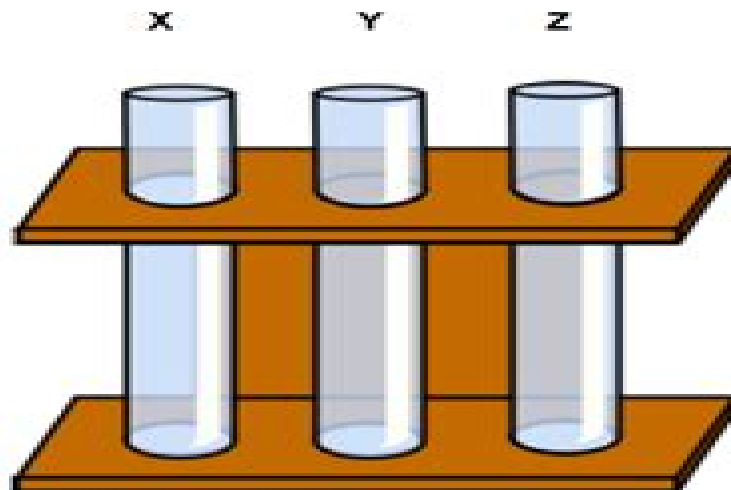


QUESTION 10

The three test tubes, X, Y and Z below, contain colourless solutions. The solutions in each test tube can either be potassium iodide or sodium carbonate or magnesium sulphate. Tests are carried out to determine which test tube contains which salts.



The following tests were carried out and the observations were made:

- Barium chloride solution is added to solution X; the solution remains colourless.
- Barium chloride solution is added to Y and Z; a white precipitate is formed in both cases.
- Dilute nitric acid is added to the precipitate in solution Y; the precipitate dissolves.

(2)

10.1 Define the term *dissociation*.

10.2 Give a reason why barium chloride solution can be used as electrolyte.

(2)

10.3 Write down the name of the precipitate in test tube Z.

(1)

Use the above information to identify the solutions in each of the test tubes X, Y and Z.

10.4 Write down the LETTER that represents the test tube and next to it, the NAME of the solution and the reason for the choice.

(6)

10.5 What type of reaction takes place between the precipitate in Y and the nitric acid?

(1)

[12]

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 - The Rise of Digital Reading Grade 10 Physical Science Exemplar For 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Physical Science Exemplar For 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Physical Science Exemplar For 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Physical Science Exemplar For 2014
 - Personalized Recommendations
 - Grade 10 Physical Science Exemplar For 2014 User Reviews and Ratings
 - Grade 10 Physical Science Exemplar For 2014 and Bestseller Lists
5. Accessing Grade 10 Physical Science Exemplar For 2014 Free and Paid eBooks
 - Grade 10 Physical Science Exemplar For 2014 Public Domain eBooks
 - Grade 10 Physical Science Exemplar For 2014 eBook Subscription Services
 - Grade 10 Physical Science Exemplar For 2014 Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
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 - Grade 10 Physical Science Exemplar For 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
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- Participating in Virtual Book Clubs
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- 9. Balancing eBooks and Physical Books Grade 10 Physical Science Exemplar For 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Physical Science Exemplar For 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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 - Setting Reading Goals Grade 10 Physical Science Exemplar For 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 10 Physical Science Exemplar For 2014
 - Fact-Checking eBook Content of Grade 10 Physical Science Exemplar For 2014
 - Distinguishing Credible Sources
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 - Exploring Educational eBooks
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