

Name _____

Date _____

CODOMINANCE & INCOMPLETE DOMINANCE PRACTICE WORKSHEET

1. Explain the difference between incomplete and codominance.

incomplete is the mix of physical traits together, codominance is both present equally

Codominance Problems

2. In a certain fish, blue scales (B) and red scales (R) are codominant. When a fish has the hybrid genotype, it has a patchwork of blue and red scales.
- What is the genotype for blue fish? BB
 - What is the genotype for red fish? RR
 - What is the genotype for patchwork fish? BR
3. What happens if you breed a patchwork fish with a fish that only has blue scales?
- What is the probability of having fish with red scales? 0 %
 - What is the probability of having fish with patchwork scales? 50 %

	B	R
B	BB	BR
B	BB	BR

4. Two patchwork fish are crossed. What is the probability that they will have patchwork fish offspring?
50 %

	B	R
B	BB	BR
R	BR	RR

5. Two short-tailed (Manx) cats are bred together. They produce three kittens with long tails, five short tails, and two without any tails. From these results, how do you think tail length in these cats are inherited? Show the genotypes for both the parents and the offspring to support your answer.

incomplete dominance, since there is a range of tails that can be made

Genetics Practice Problems Incomplete Dominance Answers

J Spring



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to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the

end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1
 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide (NaOH) and oxalic acid ($H_2C_2O_4 \cdot 2H_2O$) and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a

solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance H_2CO_3 aq 2NaOH aq Na_2CO_3 H_2O to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of khp in the

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