

General Instructions:

- This question paper consists of 39 questions in 5 sections.
- All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- Section A consists of 20 objective type questions carrying 1 mark each.
- Section B consists of 6 very short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- Section C consists of 7 short answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- Section D consists of 3 long answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

Section-A

Select and write the most appropriate option out of the four options given for each of the questions 1 - 20.

1X20= 20M

Q.1) Given the reaction:



1

Which of the following statements is true?

- Zinc is reduced and copper is oxidised.
- Zinc is oxidised and copper is reduced.
- Both zinc and copper are oxidised.
- Both zinc and copper are reduced.

Q.2) The reaction between MnO_2 with HCl is depicted in the following diagram. It was observed that a gas with bleaching abilities was released.

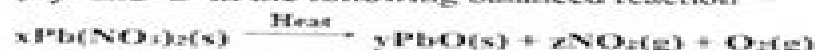
1



The chemical reaction between above mentioned reactants:

- displacement reaction
- combination reaction
- redox reaction
- decomposition reaction.

Q.3) Identify 'x', 'y' and 'z' in the following balanced reaction



1

(a) 2, 4, 2

(b) 2, 2, 4

(c) 2, 4, 4

(d) 4, 2, 2

General Paper 2 80mark Scheme

Robert R. Redfield



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Table of Contents General Paper 2 80mark Scheme

1. Understanding the eBook General Paper 2 80mark Scheme
 - The Rise of Digital Reading General Paper 2 80mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying General Paper 2 80mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Paper 2 80mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Paper 2 80mark Scheme
 - Personalized Recommendations
 - General Paper 2 80mark Scheme User Reviews and Ratings
 - General Paper 2 80mark Scheme and Bestseller Lists
5. Accessing General Paper 2 80mark Scheme Free and Paid eBooks
 - General Paper 2 80mark Scheme Public Domain eBooks
 - General Paper 2 80mark Scheme eBook Subscription Services
 - General Paper 2 80mark Scheme Budget-Friendly Options

6. Navigating General Paper 2 80mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - General Paper 2 80mark Scheme Compatibility with Devices
 - General Paper 2 80mark Scheme Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Paper 2 80mark Scheme
 - Highlighting and Note-Taking General Paper 2 80mark Scheme
 - Interactive Elements General Paper 2 80mark Scheme
8. Staying Engaged with General Paper 2 80mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Paper 2 80mark Scheme
9. Balancing eBooks and Physical Books General Paper 2 80mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Paper 2 80mark Scheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine General Paper 2 80mark Scheme
 - Setting Reading Goals General Paper 2 80mark Scheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of General Paper 2 80mark Scheme
 - Fact-Checking eBook Content of General Paper 2 80mark Scheme
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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