

## 3.240

A piston/cylinder arrangement B is connected to a 1-m<sup>3</sup> tank A by a line and valve, shown in Fig. P3.240. Initially both contain water, with A at 100 kPa, saturated vapor and B at 400°C, 300 kPa, 1 m<sup>3</sup>. The valve is now opened and, the water in both A and B comes to a uniform state.

- Find the initial mass in A and B.
- If the process results in  $T_2 = 200^\circ\text{C}$ , find the heat transfer and work.

Solution:

C.V.: A + B. This is a control mass.

$$\text{Continuity equation: } m_2 = (m_{A1} + m_{B1}) = 0;$$

$$\text{Energy: } m_2 u_2 = m_{A1} u_{A1} + m_{B1} u_{B1} = {}_1Q_2 + {}_1W_2$$

$$\text{System: if } V_B \geq 0 \text{ piston floats} \Rightarrow P_B = P_{B1} = \text{const.}$$

$$\text{if } V_B = 0 \text{ then } P_2 < P_{B1} \text{ and } v = V_A/m_{\text{tot}} \text{ see P-V diagram}$$

$${}_1W_2 = \int P_B dV_B = P_{B1}(V_2 - V_1)_B = P_{B1}(V_2 - V_1)_{\text{tot}}$$

State A1: Table B.1.1,  $x = 1$

$$v_{A1} = 1.694 \text{ m}^3/\text{kg}, \quad u_{A1} = 2506.1 \text{ kJ/kg}$$

$$m_{A1} = V_A/v_{A1} = 0.5903 \text{ kg}$$

State B1: Table B.1.2 sup. vapor

$$v_{B1} = 1.0315 \text{ m}^3/\text{kg}, \quad u_{B1} = 2965.5 \text{ kJ/kg}$$

$$m_{B1} = V_{B1}/v_{B1} = 0.9695 \text{ kg}$$

$$m_2 = m_{\text{TOT}} = 1.56 \text{ kg}$$

$$* \text{ At } (T_2, P_{B1}) \quad v_2 = 0.7163 > v_a = V_A/m_{\text{tot}} = 0.641 \text{ so } V_{B2} > 0$$

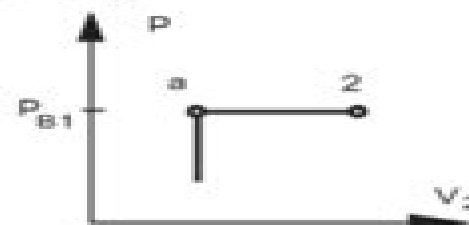
$$\text{so now state 2: } P_2 = P_{B1} = 300 \text{ kPa}, T_2 = 200^\circ\text{C}$$

$$\rightarrow u_2 = 2650.7 \text{ kJ/kg} \text{ and } V_2 = m_2 v_2 = 1.56 \times 0.7163 = 1.117 \text{ m}^3$$

$$(\text{we could also have checked } T_a \text{ at: } 300 \text{ kPa, } 0.641 \text{ m}^3/\text{kg} \rightarrow T = 155^\circ\text{C})$$

$${}_1W_2 = P_{B1}(V_2 - V_1) = -264.82 \text{ kJ}$$

$${}_1Q_2 = m_2 u_2 - m_{A1} u_{A1} - m_{B1} u_{B1} + {}_1W_2 = -484.7 \text{ kJ}$$



# Fundamentals Of Thermodynamics 8th Borgnakke Solution

**JE Gale**



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## **Table of Contents Fundamentals Of Thermodynamics 8th Borgnakke Solution**

1. Understanding the eBook Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - The Rise of Digital Reading Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Advantages of eBooks Over Traditional Books
2. Identifying Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - User-Friendly Interface
4. Exploring eBook Recommendations from Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Personalized Recommendations
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution User Reviews and Ratings
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution and Bestseller Lists
5. Accessing Fundamentals Of Thermodynamics 8th Borgnakke Solution Free and Paid eBooks

- Fundamentals Of Thermodynamics 8th Borgnakke Solution Public Domain eBooks
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution eBook Subscription Services
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution Budget-Friendly Options
6. Navigating Fundamentals Of Thermodynamics 8th Borgnakke Solution eBook Formats
- ePub, PDF, MOBI, and More
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution Compatibility with Devices
  - Fundamentals Of Thermodynamics 8th Borgnakke Solution Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Highlighting and Note-Taking Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Interactive Elements Fundamentals Of Thermodynamics 8th Borgnakke Solution
8. Staying Engaged with Fundamentals Of Thermodynamics 8th Borgnakke Solution
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Fundamentals Of Thermodynamics 8th Borgnakke Solution
9. Balancing eBooks and Physical Books Fundamentals Of Thermodynamics 8th Borgnakke Solution
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Fundamentals Of Thermodynamics 8th Borgnakke Solution
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Fundamentals Of Thermodynamics 8th Borgnakke Solution
- Setting Reading Goals Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Fundamentals Of Thermodynamics 8th Borgnakke Solution
- Fact-Checking eBook Content of Fundamentals Of Thermodynamics 8th Borgnakke Solution
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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proceso metodológico en economía 12

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