

Science 10 Final Exam Chemistry Review KEY

Part 1: Definition - Compare and contrast **THREE** of the following terms: (9marks)

1. Ionic compound and molecular compound
 - a. Ionic – A compound consisting of a metal and nonmetal, metal and polyatomic ion, or nonmetal and polyatomic ion.
 - b. Molecular – A compound consisting of 2 nonmetals
 - c. Both are created when ions combine with each other
2. Alkyl halide and Halogen
 - a. Alkyl Halide - Group 1 in the periodic table (ex/ Sodium)
 - b. Halogen - Group 17 in the Periodic Table (ex/ Fluorine)
 - c. Both are groups within the Periodic Table
3. Reactants and products
 - a. Reactant – Chemicals before any mixing has occurred
 - b. Product – Chemicals that are created after mixing has occurred
 - c. Both are involved in chemical formulae
4. Metals and nonmetals
 - a. Metals – Left side of the periodic table
 - b. Nonmetals – Right side of the periodic table
 - c. Both are found in the Periodic table and are separated by the metalloids

Part 2: Multiple Choice - Choose the **best** response for the following. (10 marks)

1. In class we watched a video about a group of chemicals that reacted explosively with water. These were:
 - a. Halogens
 - b. Alkyl Halides
 - c. Noble Gases
 - d. Earth Alkaline Metals
2. Fluorine is a(n):
 - a. Alkyl halide
 - b. Halogen
 - c. Noble gas
 - d. Transition metal
3. Combining Neon with Beryllium would create a:
 - a. Ionic Compound
 - b. Molecular compound
 - c. Neon wouldn't combine with Beryllium
 - d. Inorganic compound
4. The third shell can hold a maximum of:
 - a. 2 protons
 - b. 2 electrons
 - c. 8 protons
 - d. 8 electrons

Final Exam Grchemistry Scope

Kathleen Armour



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