

C h e m i s t r y 1 1

Atomic Theory

The Rutherford Model

To account for Rutherford's discovery of the nucleus, the Rutherford model was developed. This model shows a small, positively charged ball at the center of the atom, with most of the mass concentrated there. Electrons orbit the nucleus.

Notes

Activity 1.1: Model

Students will develop a model of the atom that is consistent with the experimental evidence.



Model 1: A simple model of an atom with a central nucleus and three orbiting electrons.

Model 2:

Model 3: A more complex model of an atom with a central nucleus and three orbiting electrons.

Model 4:

Model

Students will develop a model of the atom that is consistent with the experimental evidence.

Model 1:



Model 2: A more complex model of an atom with a central nucleus and three orbiting electrons.

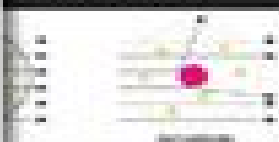
Model 3: A more complex model of an atom with a central nucleus and three orbiting electrons.



Hydrogen Atom

Model 1: A simple model of an atom with a central nucleus and three orbiting electrons.

Model 2: A more complex model of an atom with a central nucleus and three orbiting electrons.



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**Klaus Ambos-Spies, Benedikt
Löwe, Wolfgang Merkle**



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