MATTER AND CHANGE

SECTION 2.1 PROPERTIES OF MATTER (pages 39-42)

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you have to differentiate the three states of matter. It also defines a physical property and lists examples of physical properties and physical changes.

▶ Describing Matter (page 39)

- The ______ of an object is a measure of the amount of matter the object contains.
- 2. How does an exensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

▶ Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a

TO LA COSTA DA FASOR

- Is the following sentence true or false? All samples of a substance have different physical properties.
- A physical property is a quality or condition of a substance that can be observed or reasured without

changing the substance's composition.

- 6. Circle the letter of the term that is NOT a physical property.
 - m. Incombiness
- e., besilling pesint.
- by collow
- (d.) rm elting
- Look at Table 2.1 on page 40. What is the melting point of bromine? 7.1.
- Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and meks at 115°C.
 - a. wulfter
 - b. chlorine
 - e. gold
 - d. copper

Manual Chemistry Matter And Change Answer Key

Dunglison. Robley. M.D.

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