

(iii) What does  $xH_2O$  represent in the formula  $Al_2(SO_4)_3 \cdot xH_2O$ ?

Water of crystallisation.

(1)

$H_2O$  crystals to constant mass.

has a mass of 5.76g.

value of  $x$ .

+



$$\frac{5.76g}{18}$$

$$= 0.32 \text{ mol}$$

$$\frac{0.02}{16}$$

$$= 16$$

$x =$

# June 2013 Past Paper F321

**Peter Cann, Peter Hughes**



**June 2013 Past Paper F321:**

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