

Name: \_\_\_\_\_

## Honors Chem Test: Ch. 7

**Multiple Choice—1 point each**

Identify the choice that best completes the statement or answers the question.

- How many atoms of fluorine are present in a molecule of carbon tetrafluoride,  $\text{CF}_4$ ?
  - 1
  - 2
  - 4
  - 5
- Name the compound  $\text{Al}_2\text{S}_3$ .
  - aluminum sulfate
  - aluminum sulfur
  - aluminum(II) sulfate
  - aluminum sulfide
- Name the compound  $\text{CF}_4$ .
  - calcium fluoride
  - carbon fluoride
  - carbon tetrafluoride
  - mono-carbon quadrafluoride
- Name the compound  $\text{N}_2\text{O}_5$ .
  - dinitrogen oxide
  - nitrogen trioxide
  - nitric oxide
  - dinitrogen trioxide
- What is the formula for diarsenic trisulfide?
  - $\text{As}_2\text{S}_3$
  - $\text{AsS}_3$
  - $\text{As}_3\text{S}_4$
  - $\text{As}_3\text{S}_3$
- What is the formula for sulfur dichloride?
  - $\text{SCl}$
  - $\text{SCl}_2$
  - $\text{S}_2\text{Cl}$
  - $\text{S}_2\text{Cl}_2$
- What is the formula for hydrochloric acid?
  - $\text{HF}$
  - $\text{HCl}$
  - $\text{HClO}$
  - $\text{H}_2\text{CO}_3$
- The molar mass of an element is the mass of one
  - atom of the element
  - crystal of the element
  - gram of the element
  - mole of the element
- The molar mass of  $\text{MgI}_2$  is
  - the sum of the masses of 1 mol of Mg and 2 mol of I.
  - the sum of the masses of 1 mol of Mg and 1 mol of I.
  - the sum of the masses of 1 atom of Mg and 2 atoms of I.
  - the sum of the masses of 1 atom of Mg and 1 atom of I.

# Honors Chemistry Chapter Practice Test

**Daniel F McAuley**



## Honors Chemistry Chapter Practice Test:

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## **Table of Contents Honors Chemistry Chapter Practice Test**

1. Understanding the eBook Honors Chemistry Chapter Practice Test
  - The Rise of Digital Reading Honors Chemistry Chapter Practice Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Honors Chemistry Chapter Practice Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Honors Chemistry Chapter Practice Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Honors Chemistry Chapter Practice Test
  - Personalized Recommendations
  - Honors Chemistry Chapter Practice Test User Reviews and Ratings
  - Honors Chemistry Chapter Practice Test and Bestseller Lists
5. Accessing Honors Chemistry Chapter Practice Test Free and Paid eBooks
  - Honors Chemistry Chapter Practice Test Public Domain eBooks
  - Honors Chemistry Chapter Practice Test eBook Subscription Services
  - Honors Chemistry Chapter Practice Test Budget-Friendly Options

6. Navigating Honors Chemistry Chapter Practice Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Honors Chemistry Chapter Practice Test Compatibility with Devices
  - Honors Chemistry Chapter Practice Test Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Honors Chemistry Chapter Practice Test
  - Highlighting and Note-Taking Honors Chemistry Chapter Practice Test
  - Interactive Elements Honors Chemistry Chapter Practice Test
8. Staying Engaged with Honors Chemistry Chapter Practice Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Honors Chemistry Chapter Practice Test
9. Balancing eBooks and Physical Books Honors Chemistry Chapter Practice Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Honors Chemistry Chapter Practice Test
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Honors Chemistry Chapter Practice Test
  - Setting Reading Goals Honors Chemistry Chapter Practice Test
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Honors Chemistry Chapter Practice Test
  - Fact-Checking eBook Content of Honors Chemistry Chapter Practice Test
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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