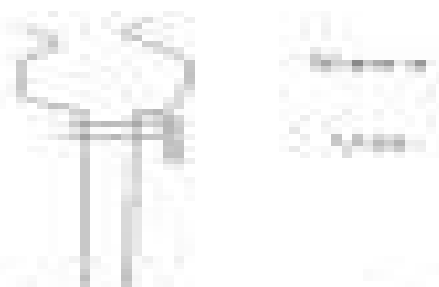


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (4pts)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of this solution, consistently measured and 4 conditions

(a) Temperature appropriate method of measurement and result obtained in the in solution the solution

(b) What is reason for the existence of your answer

(c) Describe how the pH of the solution changes and the pH of the solution (4pts) (4pts) (4pts) (4pts)

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