

NAME \_\_\_\_\_ INDEX NUMBER \_\_\_\_\_

SCHOOL \_\_\_\_\_ DATE \_\_\_\_\_

## SIMPLE CLASSIFICATION OF SUBSTANCES

1. 1995 Q 15

Explain how you would separate a mixture of nitrogen and oxygen gases given that their boiling points are  $-196^{\circ}\text{C}$  and  $-183^{\circ}\text{C}$  respectively. (2 marks)

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.....

2. 1995 Q 8 P2

(a) The table below gives information about the major constituents of crude oil. Study it and answer the questions that follow.

Constituent	Boiling point ( $^{\circ}\text{C}$ )
Gases	Below $-90$
Petrol	$40-175$
Kerosene	$175-250$
Diesel oil	$250-350$
Lubricating oil	$350-400$
Bitumen	Above $400$

(i) Which one of the constituents of crude oil has molecules with the highest number of carbon atoms? (2 marks)

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(ii) Name the process you would use to separate a mixture of petrol and diesel and explain how the separation takes place. (2 marks)

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(iii) Explain why the constituent of crude oil and write its formula. (1 mark)

# Form 1 Chemistry Questions

**Career Point Kota**



## **Form 1 Chemistry Questions:**

The book delves into Form 1 Chemistry Questions. Form 1 Chemistry Questions is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Form 1 Chemistry Questions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to Form 1 Chemistry Questions
  - Chapter 2: Essential Elements of Form 1 Chemistry Questions
  - Chapter 3: Form 1 Chemistry Questions in Everyday Life
  - Chapter 4: Form 1 Chemistry Questions in Specific Contexts
  - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Form 1 Chemistry Questions. This chapter will explore what Form 1 Chemistry Questions is, why Form 1 Chemistry Questions is vital, and how to effectively learn about Form 1 Chemistry Questions.
3. In chapter 2, this book will delve into the foundational concepts of Form 1 Chemistry Questions. This chapter will elucidate the essential principles that must be understood to grasp Form 1 Chemistry Questions in its entirety.
4. In chapter 3, this book will examine the practical applications of Form 1 Chemistry Questions in daily life. The third chapter will showcase real-world examples of how Form 1 Chemistry Questions can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Form 1 Chemistry Questions in specific contexts. This chapter will explore how Form 1 Chemistry Questions is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Form 1 Chemistry Questions. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Form 1 Chemistry Questions.

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