

# Chemistry Form 5 Chapter 1

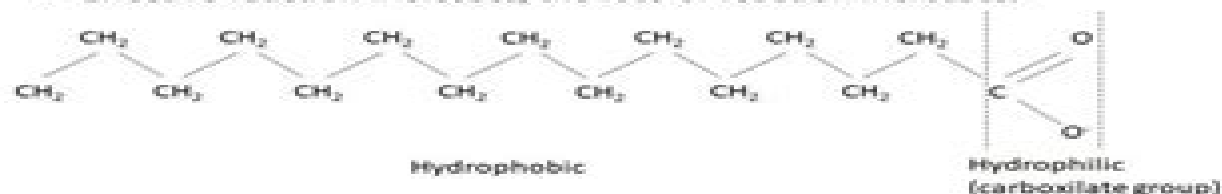
## Rate of Reaction

- Rate of reaction is the change in quantity of reactant or product against time
- Method used to measure the volume of hydrogen gas is **water displacement method**

<u>Fast reaction</u>	<u>Slow reaction</u>
Short time taken for the reaction	Long time taken for the reaction
High rate of reaction	Low rate of reaction
Sodium + Water	Copper(II) oxide + Hydrogen gas
Neutralisation	Fermentation
Precipitation	Rusting

## Effective reaction

- Effective reaction are **collisions** between the **particles of reactants** that cause a reaction
- Occur when the particles of reactants have energy that is equal or higher than **activation energy**.
- Effective reaction increases, the rate of reaction increases.



## Graph

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