

### Question 7

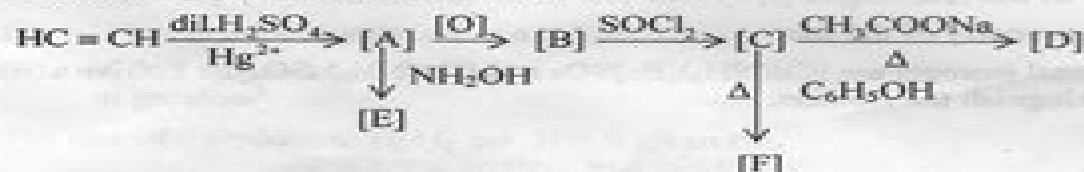
- (a) In the extraction of zinc from zinc blende: [2]
- (i) Give an equation to show how zinc oxide is converted to zinc.
  - (ii) How is impure zinc finally electro-refined?
- (b) Explain why: [3]
- (i) Transition elements form coloured compounds.
  - (ii) Interhalogen compounds are more reactive than their constituent elements.
  - (iii)  $\text{Cu}^+$  is diamagnetic but  $\text{Cu}^{2+}$  is paramagnetic. ( $Z = 29$ )

## SECTION C

Answer any two questions.

### Question 8

- (a) How can the following conversions be brought about:
- (i) Nitro benzene to benzene diazoniumchloride. [3]
- (ii) Propanoic acid to ethylamine. [2]
- (iii) Benzoic acid to benzaldehyde. [2]
- (b) Identify the compounds A, B, C, D, E and F: [3]



### Question 9

- (a) Write balanced chemical equations for the following reactions and name the reactions: [3]  
(i) Acetamide is heated with bromine and sodium hydroxide solution.  
(ii) Benzaldehyde is treated with 50% sodium hydroxide solution.
- (b) Give one chemical test to distinguish between the following pairs of compounds: [3]  
(i) Acetone and phenol.  
(ii) Formic acid and Acetic acid.

# Form 4 Chemistry Question Paper 2013

**AW Chickering**



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