Acids and Bases

Question 1:		Choose the correct answer for each of the following questions:				
1.1	The element present in all acids					
	A	oxygen	В	carbon o	dioxide	
	C	nitrogen	D	hydre		
1.2	A sample of an acid in a beaker has a pH of 4. An alka.					pH of
	the liquid in the beaker?					
	A	decreases	В	stays the	e same	
	C	Increases	D	neutral		
1.3	Choose the incorrect combination:					
	A	HNH₄OH – Nitric acid.	В	H ₂ SO	sulphuric ac	
	C	HCl - Hydrochloric acid.	D		rbonic acid.	
1.4	The colour of blue litmus paper in an acid is					
	A	yellow	В	en		
	C	blue				
1.5	Which substance is always produced in eaction. ochloric and and lithium?					
	A	water		hydrogen 8-		
	C	oxygen gas		on dioxide gas		
1.6	Which:	substance is always produ in	the rea.		n hydrochloric acid and sodium	
	hydroxi	ide?				
	A	water		hydrogen gas		
	C	oxygen gas		ecipitate		
1.7	Which formula repr					
	A	кон		KCI		
	C	СН₃ОН		CH₃COOH		
1.8	T	od of adding an a	irop b	y drop:		
		neutralisation	В	precipita	ation	
		eva-	D	decomp	osition	[8]
Quest.	.ne genera reaction for an			acid with	the following compounds:	
2.1	Base:					
2.2	Carbonate:					
2.3	Metal:			-(8)-5	turdvehamo-co.	za

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- 1. Understanding the eBook Finalexam 2013 Natural Science Grade 9
 - o The Rise of Digital Reading Finalexam 2013 Natural Science Grade 9
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Finalexam 2013 Natural Science Grade 9
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Finalexam 2013 Natural Science Grade 9
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Finalexam 2013 Natural Science Grade 9
 - Personalized Recommendations
 - Finalexam 2013 Natural Science Grade 9 User Reviews and Ratings
 - Finalexam 2013 Natural Science Grade 9 and Bestseller Lists
- 5. Accessing Finalexam 2013 Natural Science Grade 9 Free and Paid eBooks

- Finalexam 2013 Natural Science Grade 9 Public Domain eBooks
- Finalexam 2013 Natural Science Grade 9 eBook Subscription Services
- Finalexam 2013 Natural Science Grade 9 Budget-Friendly Options
- 6. Navigating Finalexam 2013 Natural Science Grade 9 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Finalexam 2013 Natural Science Grade 9 Compatibility with Devices
 - Finalexam 2013 Natural Science Grade 9 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Finalexam 2013 Natural Science Grade 9
 - Highlighting and Note-Taking Finalexam 2013 Natural Science Grade 9
 - Interactive Elements Finalexam 2013 Natural Science Grade 9
- 8. Staying Engaged with Finalexam 2013 Natural Science Grade 9
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Finalexam 2013 Natural Science Grade 9
- 9. Balancing eBooks and Physical Books Finalexam 2013 Natural Science Grade 9
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Finalexam 2013 Natural Science Grade 9
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Finalexam 2013 Natural Science Grade 9
 - Setting Reading Goals Finalexam 2013 Natural Science Grade 9
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Finalexam 2013 Natural Science Grade 9
 - Fact-Checking eBook Content of Finalexam 2013 Natural Science Grade 9
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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