

Answers to Topic questions

Topic 1 Basic agricultural chemistry Answers

- 1
 - a Molecules can be broken down chemically into the elements of which they are composed. Example: carbon dioxide, water, sodium chloride, etc.
 - b Isotopes: atoms of the same element with different atomic weights. Example: chlorine.
 - c Valence electrons: outermost electrons around an atom.
- 2 The atom has a small, positively charged core called the nucleus. The atom is electrically neutral and consists of: 1) electrons – negatively charged particles contained in a large outer volume called an electron cloud; 2) protons – positively charged particles concentrated in a small inner volume that contains most of the mass of the atom; 3) neutrons – electrically neutral and occur in the nucleus.
- 3
 - a alkali-earth metal
 - b fluorine / chlorine / bromine / iodine
- 4
 - a covalent bonding
 - b oxidation number
 - c chemical formula
 - d polar
 - e hydrogen bonding
- 5 b.
- 6 Hydrogen bonds that hold water molecules together are quite strong. But covalent bonds between atoms in a water molecule are stronger. Thus hydrogen bonds often break and reform which provide water's ability to flow.
- 7
 - a carbon, oxygen
 - b 2 of Fe, Mn, B, Mo, Cu, Zn, Cl, Co
- 8 Plants use CO_2 and H_2O to build complex organic compounds. They are excreted by animals during respiration.
- 9
 - a alcohol
 - b aldehyde
 - c amide
 - d sugar

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K Payea



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