

FIFTY FREQUENTLY FORGOTTEN FUN FACTS

This packet contains topics from each of the units we worked on this year with questions. Most of the questions are similar to what you would expect to see on Part B2 and C of the Regents Exam in Chemistry. The multiple choice questions mirror common questions found on Parts A and B1.

I. ATOMIC STRUCTURE & NUCLEAR CHEMISTRY

1) Protons are +1 each with a mass of 1 amu each, the number of protons = atomic number, nuclear charge = + (# protons). [Periodic Table]

- a) How many protons are there in a nucleus of Kr-85? _____
- b) What is the nuclear charge of an atom of Br? _____
- c) What is the mass of the protons in a nucleus of O-15? _____

2) Neutrons are neutral with a mass of 1 amu each, # neutrons = mass # - atomic number. Isotopes = atoms of the same element (same atomic #) but different # of neutrons (mass #). [Periodic Table]

- a) How many neutrons are there in the nucleus of $^{56}_{26}\text{Fe}$? _____

- b) Circle the two nuclei that are isotopes of each other: $^{16}_8\text{O}$ $^{15}_7\text{N}$ $^{16}_8\text{O}$ $^{19}_9\text{F}$

3) Electrons are each -1 with a mass that is VERY, VERY tiny compared to the mass of a proton or neutron.

- a) Which particle has a mass that is 1/1836th the mass of a proton?

- 1) ^4_2He 2) ^1_1H 3) $^{0}_{-1}\text{e}$ 4) ^1_0n

4) Natural Decay: Parent Nuclide → Decay particle + daughter nuclide [Tables N and O]

- a) Write the decay for U-238: _____
- b) Write the decay for K-37: _____
- c) Write the decay for P-32: _____

5) Artificial Transmutation is when a relatively stable nucleus is impacted by a particle bullet at high speeds and becomes an unstable nucleus of a different element. Nuclear fission occurs when nuclei of U-235 or Pu-239 are impacted by a neutron and split into two smaller nuclei and more neutrons. Nuclear fusion occurs when two small nuclei of hydrogen combine at high temperatures and pressures to form larger nuclei of helium. Both fission and fusion convert mass into a huge amount of energy.

Given the nuclear reactions:

- 1) $^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{92}_{36}\text{Kr} + ^{141}_{56}\text{Ba} + 3 ^1_0\text{n}$ 2) $^{239}_{94}\text{Pu} + ^4_2\text{He} \rightarrow ^{242}_{96}\text{Cm} + ^1_0\text{n}$
- 3) $^{226}_{88}\text{Ra} \rightarrow ^{222}_{86}\text{Rn} + ^4_2\text{He}$ 4) $^2_1\text{H} + ^2_1\text{H} \rightarrow ^4_2\text{He}$

- a) Which reaction represents natural decay? _____
- b) Which reaction represents artificial transmutation? _____
- c) Which reaction represents nuclear fission? _____
- d) Which reaction represents nuclear fusion? _____

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Zhe Xu



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Table of Contents Fifty Frequently Forgotten Fun Facts Chemistry Answers

1. Understanding the eBook Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - The Rise of Digital Reading Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Personalized Recommendations
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers User Reviews and Ratings

- Fifty Frequently Forgotten Fun Facts Chemistry Answers and Bestseller Lists
- 5. Accessing Fifty Frequently Forgotten Fun Facts Chemistry Answers Free and Paid eBooks
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers Public Domain eBooks
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers eBook Subscription Services
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers Budget-Friendly Options
- 6. Navigating Fifty Frequently Forgotten Fun Facts Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers Compatibility with Devices
 - Fifty Frequently Forgotten Fun Facts Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Highlighting and Note-Taking Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Interactive Elements Fifty Frequently Forgotten Fun Facts Chemistry Answers
- 8. Staying Engaged with Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Fifty Frequently Forgotten Fun Facts Chemistry Answers
- 9. Balancing eBooks and Physical Books Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Fifty Frequently Forgotten Fun Facts Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Setting Reading Goals Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Fact-Checking eBook Content of Fifty Frequently Forgotten Fun Facts Chemistry Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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