

Chemistry 12 - Review of Significant Digits

TOTAL 28

The rules for zeros in significant digits are as follows:

- All zeros between non-zero digits are significant.
  - Zeros at the beginning of a number ( eg. 0.0095) are **NOT SIGNIFICANT** !  
If the number 0.0095 was written in scientific notation, it would be:  $9.5 \times 10^{-3}$ . The exponent is not counted as significant so this number has 2 significant digits.
  - Zeros on the right side of a number (at the end) are significant if the **DECIMAL POINT** is shown.  
  
eg) 50.00 has 4 significant digits  
43.0 has 3 significant digits  
20. has 2 significant digits  
100. has 3 significant digits
  - Zeros to the left of an **UNDERSTOOD** decimal point are **NOT** significant.  
  
eg) 300 has 1 significant digit  
10 000 has 1 significant digit  
12 320 has 4 significant digits  
420 has 2 significant digits
1. Find the number of significant digits in each of the following measurements:

- |                                    |                                     |
|------------------------------------|-------------------------------------|
| a) 3.4005 ..... (5)                | f) $9.080 \times 10^{-3}$ ..... (4) |
| b) 2580 ..... (3)                  | g) 1.00 ..... (3)                   |
| c) $3.20 \times 10^{-2}$ ..... (3) | h) 0.0027890 ..... (5)              |
| d) 0.000308 ..... (3)              | i) 320 000 ..... (2)                |
| e) 23.000 ..... (5)                | j) 9 ..... (1)                      |

# Instructional Fair Inc Chemistry If8766 Answer Key

**Guillaume Favre**



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