

Graphing Periodic Trends Lab

Name _____ Hr _____

PURPOSE: To determine how certain properties are periodic when the elements are arranged in periods or groups.

DISCUSSION: There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

Ionization energy - The energy required to remove an electron from an atom.

DATA

Properties of the First 18 Elements

Element	Atomic Number	Atomic Radius (Angstroms)	Ionization Energy
H	1	0.79	314
He	2	0.49	567
Li	3	2.05	124
Be	4	1.4	215
B	5	1.17	191
C	6	0.91	260
N	7	0.75	335
O	8	0.65	314
F	9	0.57	402
Ne	10	?	?
Na	11	2.23	119
Mg	12	1.72	176
Al	13	1.82	138
Si	14	1.46	168
P	15	1.23	242
S	16	1.09	239
Cl	17	0.97	299
Ar	18	?	?

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with question marks.

1. Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)
2. Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)

Graphing Trends In The Periodic Table Lab Answers

N Colangelo



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Prentice Hall Science Explorer Michael J. Padilla, 2002 **The Science Teachers Bulletin**, 1989 General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems *Laboratory Manual* Jo A. Beran, 1990 **Prentice Hall Chemistry**, 2000 **Hands-On General Science Activities With Real-Life Applications** Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5-12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life *Exploring Physical Science in the Laboratory* John T. Salinas, 2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging *Exploring Physical Science in the Laboratory* guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts **Higher Chemistry: Practice Papers for SQA Exams** Barry McBride, 2017-10-30 Practise for your SQA exams with three specially commissioned Hodder Gibson Practice Exam Papers Practise with model papers written and checked by experienced markers and examiners Get extra advice with specially written study skills guidance sections Gain vital extra marks and avoid common mistakes with examiner tips **A World From Dust** Ben McFarland, 2016-03-07 *A World From Dust* describes how a set of chemical rules combined with the principles of evolution in order to create an environment in which life as we know it could unfold Beginning with simple mathematics these predictable rules led to the advent of the planet itself as well as cells organs and organelles ecosystems and increasingly complex life forms McFarland provides an accessible discussion of a geological history as well describing how the inorganic matter on Earth underwent chemical reactions with air and water allowing for life to emerge from the world's first rocks He traces the history of life all the way to modern neuroscience and shows how the bioelectric signals that make up the human brain were formed Most popular science books on the topic present either the physics of how the universe formed or the biology of how complex life came about this book's approach would be novel in that it condenses in an engaging way the chemistry that links the two fields This book is an accessible and multidisciplinary look at how life on our planet came to be and how it continues to develop and change even today This book includes 40 illustrations by Gala Bent print artist and studio faculty member at Cornish College of the Arts and Mary Anderson medical

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