# **Graphing Periodic Trends Worksheet**

 Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

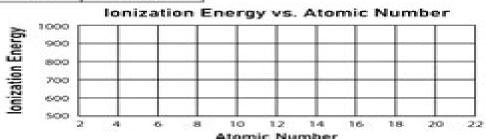
			Period 2
Element	Atomic Number	IE (kJ/mol)	
Li	3	519	
Be	4	900	
В	5	799	
C	6	1088	

Element	Atomic Number	IE (kJ/mol)		
Li	3	519		
Be	4	900		
В	5	799		
C	6	1088		

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1000	+		-	-	-		-	
500	+	-		-	-	-	-	
0						8	- 0	10

- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
- Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A				
Element	Atomic Number	IE (kJ/mol) 900		
Be	4			
Mg	12	736		
Ca	20	590		



# **Graphing Activities Chemistry Answer Key**

**Eric Vieil** 

**Graphing Activities Chemistry Answer Key:** 

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