

Molarity Calculations

Name _____	Section _____	Date _____
Molarity Calculations		
Directions: Solve each problem for the indicated quantity.		
1. What is the molarity of 1.32 L of a solution of 0.3087 moles of KMnO_4 ?		
2. How many moles of NaOH are present in 2.36 L of a 1.871 M aqueous solution of NaOH ?		
3. What volume in Liters is an aqueous solution of HCl if it has a molarity of 0.850 M and contains 0.580 moles of HCl ?		
4. What is the molarity of 0.1000 moles of KI in 31.7 mL of aqueous solution?		
5. What is the volume in mL of 1.5000 moles of NaClO with a molarity of 0.6341 M?		
6. What is the molarity of an aqueous solution of 2.35 grams of NaCl in a solution with a volume of 1.914 L?		
7. What is the volume of an aqueous solution in mL if it has a molarity of 0.112 M and contains 49.08 grams of dissolved KBr ?		
8. How many grams of H_2CO_3 are dissolved in 2.32 L of an aqueous solution with a molarity of 0.7542 M?		
9. Find the volume of an aqueous solution of KNO_3 in L if it has a molarity of 1.89 M and contains 16.27 grams of KNO_3 . If the solution contains L of water, what is the volume of the dissolved KNO_3 ?		
10. Potassium sulfate has a density of 2.66 $\frac{\text{g}}{\text{mL}}$. What is the molarity of a solution of 7.337 grams of K_2SO_4 and 1.375 L of water?		

$$\text{Molarity} = \frac{\text{Moles of Solute}}{\text{Liters of Solution}}$$

$$M = \frac{\text{mol}}{L}$$

A CHEMISTRY WORKSHEET

10 PROBLEMS

WORKED ANSWER KEY INCLUDED

Grams And Liters Molarity Calculations Answers

B Lingard



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Clinical Chemistry: Principles, Techniques, and Correlations with Navigate Advantage Access Michael L. Bishop, Edward P. Fody, Carleen Van Siclen, James March Mistler, 2022-03-10 This edition carries on the tradition of excellence for this book. If you are learning clinical chemistry or a practitioner wanting a contemporary refresher, this book is for you. Get it! Valerie L. Ng, PhD, MD, Alameda County Medical Center and Highland Hospital, Score 97.5 Stars. Clinical Chemistry: Principles, Techniques, and Correlations, Ninth Edition, is the most student-friendly clinical chemistry text available today. The Ninth Edition keeps students at the forefront of what continues to be one of the most rapidly advancing areas of laboratory medicine with clear explanations that balance analytic principles, techniques, and correlation of results with coverage of disease states. The book not only demonstrates the how of clinical testing but also the what, why, and when of testing correlations to help students develop the knowledge and interpretive and analytic skills they'll need in their future careers. The Ninth Edition's content is mapped to ASCLS entry-level curriculum and ASCP Board of Certification guidelines. Every new print copy includes Navigate Advantage Access that unlocks an interactive eBook with Knowledge Check questions and quizzes, case studies, review questions, flashcards, reference range table, general reference tables, and a supplementary chapter, Molecular Theory and Techniques. Over 80 new Case Studies, which include scenarios, lab results, and questions, give you an opportunity to apply content to clinical practice. Coverage of the latest equipment and technologies used in the modern lab prepares you for real-world practice. Practical, clinically based coverage reflects the most recent or commonly performed techniques in the clinical chemistry laboratory. Insightful coverage of the impact of problem-solving, quality assurance, and cost effectiveness on the laboratory professional prepares you for clinical practice. Useful in-text learning aids include chapter outlines and chapter objectives tables that condense and augment theory coverage and end-of-chapter questions that help you assess your level of mastery. A robust Health Professions Basic Math Review module provided in the online component provides study tools and worksheets to help you review the math concepts required to be successful. 2023, 736 pages.

Clinical Chemistry: Principles, Techniques, and Correlations, Enhanced Edition Michael L. Bishop, 2020-06-11 Clinical Chemistry: Principles, Techniques, and Correlations, Enhanced Eighth Edition, demonstrates the how, what, why, and when of clinical testing and testing correlations to help you develop the interpretive and analytic skills you'll need in your future career.

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