



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

Duration: 120
Time: 1 hour



The question paper consists of 12 pages, including 4 extra sheets.
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QUESTION 8

PHYSICAL SCIENCES PAPER 2

ACID AND BASES

GRADE 11

NOVEMBER 2020

QUESTION 8: ACID AND BASES (100 marks)

8.1. Draw a potential energy diagram for the reaction represented by

The following equation shown on the page:



QUESTION 8 (Start on a NEW page)

8.2. Sulfuric acid reacts with an organic compound with the following percentage composition:

C: 55.4%, H: 9.09%, O: 35.5%

The molar mass of the compound is 116 g mol⁻¹.

8.2.1. Calculate the molar mass of the compound. (10)

8.2.2. Determine the structure of the compound. (10)

8.2.3. Draw the structural formula. (10)

8.3. Concentrated sulfuric acid reacts with sodium hydroxide solution to form sodium sulfate. 2 g of sodium hydroxide (NaOH) is a substance that can release water in a process called a reaction after heating.

8.3.1. Define concentration in words. (10)

8.3.2. Calculate the concentration of the sodium hydroxide solution. (10)

8.4. Sodium sulfate (Na₂SO₄) is used in car airbags. For the airbag to inflate the following reaction must take place:



Calculate the volume of nitrogen gas (N₂) that would be produced at STP if 60 g of sodium azide reacts completely. (10)

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