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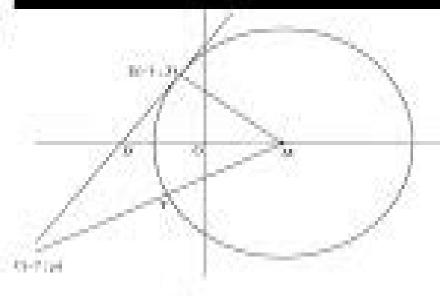
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web what is the mole fraction of licl in the solution $11\ 8$ practice problem acetone c $3\ h$ 6 o has a density of $0\ 791\ g$ ml and xylene c $8\ h$ 10 has a density of $0\ 864\ g$ ml a solution was prepared by mixing acetone and xylene calculate the mole fraction of acetone in a solution if $19\ 0$ ml of acetone was dissolved in $98\ 4$ ml of xylene $7\ 1$

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web since the mole fraction of the o 2 is 0 050 it contributes 0 050 or if you will 5 0 of the total pressure thus total pressure is 0 21 0 050 4 2 atm problem 9 a sample of 1 43 g of helium and an unweighed quantity of o 2 are mixed in a flask at room temperature