



GRADE 10 PHYSICAL SCIENCES

NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

SECTION A

QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION/NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B Na_2S
- C NaS_2
- D Na_2S_2

1.4 Strongest intermolecular force found in water?

- A ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a

$1s^2$	2s
$2s^2 2p^4$	3s
$3s^2$	3p
$3d^5$	4s

- A noble gas
- B halogen
- C alkali earth metal
- D alkali metal

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**Hugh Mac Rae, Kedir Ibrahim, Malcolm
James**



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Table of Contents Grade 10 Physical Science Exemplar 2013 November

1. Understanding the eBook Grade 10 Physical Science Exemplar 2013 November
 - The Rise of Digital Reading Grade 10 Physical Science Exemplar 2013 November
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Physical Science Exemplar 2013 November
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Physical Science Exemplar 2013 November
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Physical Science Exemplar 2013 November
 - Personalized Recommendations
 - Grade 10 Physical Science Exemplar 2013 November User Reviews and Ratings
 - Grade 10 Physical Science Exemplar 2013 November and Bestseller Lists

5. Accessing Grade 10 Physical Science Exemplar 2013 November Free and Paid eBooks
 - Grade 10 Physical Science Exemplar 2013 November Public Domain eBooks
 - Grade 10 Physical Science Exemplar 2013 November eBook Subscription Services
 - Grade 10 Physical Science Exemplar 2013 November Budget-Friendly Options
6. Navigating Grade 10 Physical Science Exemplar 2013 November eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 10 Physical Science Exemplar 2013 November Compatibility with Devices
 - Grade 10 Physical Science Exemplar 2013 November Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 10 Physical Science Exemplar 2013 November
 - Highlighting and Note-Taking Grade 10 Physical Science Exemplar 2013 November
 - Interactive Elements Grade 10 Physical Science Exemplar 2013 November
8. Staying Engaged with Grade 10 Physical Science Exemplar 2013 November
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Physical Science Exemplar 2013 November
9. Balancing eBooks and Physical Books Grade 10 Physical Science Exemplar 2013 November
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Physical Science Exemplar 2013 November
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade 10 Physical Science Exemplar 2013 November
 - Setting Reading Goals Grade 10 Physical Science Exemplar 2013 November
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 10 Physical Science Exemplar 2013 November
 - Fact-Checking eBook Content of Grade 10 Physical Science Exemplar 2013 November
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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