

- a. electron carrier
- b. high-energy sugar
- c. photosynthetic pigment
- d. pigment

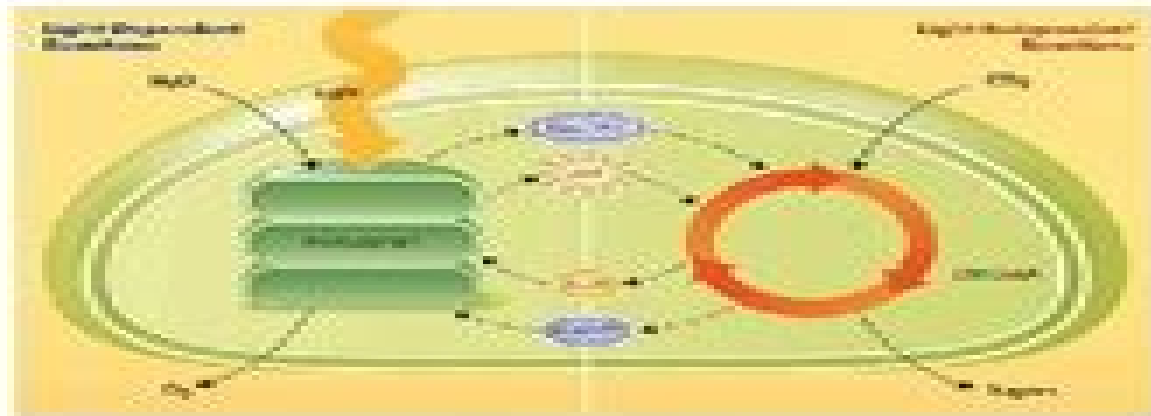


Figure 6-3

- 22. Which chemical shown in Figure 6-3 is an electron carrier molecule?
 - a. H_2O
 - b. carbon dioxide
 - c. NADP⁺
 - d. oxygen
- 23. What makes certain molecules good electron carriers?
 - a. They can accept electrons and transfer most of their energy to another molecule.
 - b. They are very large molecules, so they have lots of room to carry many electrons.
 - c. They can absorb sunlight, which is where all of the high-energy electrons come from.
 - d. They are carbohydrates and have a lot of energy which allows them to carry electrons.
- 24. Why are electron carriers needed for transporting electrons from one part of the chloroplast to another?
 - a. High-energy electrons would be destroyed.
 - b. High-energy electrons are highly reactive.
 - c. High-energy electrons are not soluble in cytoplasm.
 - d. High-energy electrons get their energy from electron carriers.
- 25. A student is collecting the gas given off from a plant in bright sunlight at a temperature of 27°C. The gas being collected is probably
 - a. carbon dioxide
 - b. glucose
 - c. oxygen
 - d. sucrose

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**Kenneth Raymond Miller, Prentice Hall
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