

Integrated Science I

Chapter 6 Section 1 & 2 Quiz Review

1. Give a real world example of a compound.
Table salt, baking soda
2. What holds atoms and ions together?
A chemical bond
3. Identify what the following types of bond models are used for:
 - a. Ball-and-stick – *shows bond angle and bond length*
 - b. Space-filling – *shows space occupied by atoms*
 - c. Structural Formula – *shows symbols of elements used*
4. Strong bonds cause atoms to have what characteristics?
Strong attraction causes high melting and boiling points of compounds
5. How does the attraction between molecules compare to the attraction between ions?
The attraction between molecules is weaker. The bond between hydrogen atoms is weaker than the bond between the hydrogen molecule and oxygen atom in water.
6. Salt has a higher melting point than sugar because of what characteristic?
Salt is formed by a ionic bond and sugar is formed by a covalent bond
7. Why would a compound with only two atoms not have a bond angle?
Bond angles form when there is two or more bonds to the same atom.
8. Why do atoms form bonds? *Bonds are formed to create compounds and to stabilize atoms that do not have a full valence electron level.*
9. Identify what types of atoms make the following types of bonds:
 - a. Metallic metal + metal
 - b. Ionic metal + non-metal
 - c. Covalent non-metal + non-metal
10. What are covalent bonds and how do these bonds affect the boiling point of a substance?
Covalent bonds form when atoms share electrons. This type of bond causes the boiling and melting points of substances to be lower.
11. What are ionic bonds and how do these bonds affect the boiling point of a substance?
Ionic bonds form when atoms gain or lose electrons. This type of bond causes the boiling and melting points of substances to be higher.
12. What affect do metallic bonds have on the properties of metals? *Metals are able to be more flexible since the electrons and atoms can move more freely. It gives metals properties such as being malleable and ductile.*

Integrated Science Chapter 19 Review Answers

Rosina Ehmann



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