

# Properties of Atoms and the Periodic Table

## Section 2 Masses of Atoms

**Preview** Section 2 of your book, using the checklist below.

- Read all section titles.
- Read all boldfaced words.
- Read all charts and graphs.
- Look at all the pictures and read their captions.
- Think about what you already know about masses of atoms.

*Write three facts you learned.*

1. Accept all reasonable answers. Protons identify an element.
2. Radioactive isotopes are used to measure the age of fossils.
3. An electron's mass is negligible.

### Review Vocabulary

mass

**Define** mass to show its scientific meaning.

amount of matter in an object

### New Vocabulary

atomic number

*Use your book or dictionary to define the following key terms.*

a number equal to the number of protons in an atom's nucleus

mass number

sum of the number of protons and neutrons in an atom

isotope

atoms of the same element with different numbers of neutrons

average atomic mass

the average mass of the mixture of isotopes of an element

### Academic Vocabulary

define

*Use a dictionary to find the scientific meaning of define.*

to determine or identify the essential qualities or meaning of something

# Instructional Fair Inc Chemistry If8766 Atomic Structure

**CO Houle**

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