

Stoichiometry - Gram Conversions

1. Use the following balanced equation to answer the questions below:



- How many grams of O_2 are produced when 34.5g of HgO decompose?
 - How many grams of HgO need to decompose to produce 0.277g of Hg ?
 - If 52.1g of HgO decomposes, then how many grams of Hg will be made?
2. When catalyzed by sulfuric acid, potassium chlorate and sucrose react to produce heat, CO_2 , H_2O , and KCl . Below is a balanced equation for this reaction.



- How many grams of CO_2 will form if 52.9g of sucrose react completely with KClO_3 ?
 - How many grams of KClO_3 are needed to completely react with 781g of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$?
 - How many moles of water will be produced if 99.1g of KClO_3 reacts completely with sucrose?
3. Nitrogen dioxide reacts with water to produce nitric acid and nitric oxide. Below is the balanced equation for this reaction.



- How many grams of NO will be made if 5.57g of NO_2 reacts completely with water?

Gram To Gram Stoichiometry Worksheet Answers

Robert R. Redfield



Gram To Gram Stoichiometry Worksheet Answers:

Gram To Gram Stoichiometry Worksheet Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Gram To Gram Stoichiometry Worksheet Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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