

Compounds

Question 1: Choose the correct letter for each question:

- 1.1 Group I is known as
A transition metals
B alkali earth metals
C alkali metals
D earth metals
- 1.2 Group II is known as
A transition metals
B alkaline earth metals
C alkali metals
D earth metals
- 1.3 Group VII is known as
A noble gases
B halogens
C halogen gases
D noble gases
- 1.4 Group VIII is known as
A non-metals
B noble gases
C halogen gases
D noble gases
- 1.5 Which of the following elements has one electron in its outer orbital?
A Phosphorous
B Gallium
C Strontium
D Barium
- 1.6 Which element is the odd one out according to the periodic table?
A Nitrogen
B Antimony
C Arsenic
D Bismuth
- 1.7 Identify the element in group 13 and period 5.
A Germanium
B Aluminium
C Tin
D Arsenic
- 1.8 Identify the element in group 14 and period 4.
A Calcium
B Strontium
C Carbon
D Silicon
- 1.9 Which of the following is an example of a chemical property of elements?
A Conductivity
B Density
C Heat capacity
D Reactivity

Grade9 Question Paper2014 Natural Science Term1

Terri Raymond



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Grade9 Question Paper2014 Natural Science Term1**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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