

PHYSICAL SCIENCES (CHEMISTRY) GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2 VERIFICATION OF BOYLE'S LAW MARKING GUIDELINES

| 1.1 • Aim stated correctly. < | |
|-------------------------------|--|
|-------------------------------|--|

e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

67.00

COOK.

- 1.2 . All apparatus listed v
 - Blovée's Apparatus
 - 2. Pump
 - All steps of the method correct and logically stated.
 correct tense used.

49.40

- The tap on the Boyle's Law apparatus was opened and then the pump was connected.
- The apparatus was pumped to the pressure.
- The oil was allowed to settle for a few minutes.
- A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
- The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
- 5 different readings were recorded in the table.

2. Observations and Recording of results:

Learners will use results obtained during experiment to complete the table below.

| Volume (V) cm ³ | Pressure (p) kPa | P.V. (*) |
|----------------------------|------------------|----------|
| | | |
| | | |
| | | |

Marking guidelines:

- Ior correct values of the volume
- If for correct values of the pressure.
- ✓✓ for correct values of

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Edwin Black

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