

















Types of Chemical Reactions

Do atoms rearrange in predictable patterns during chemical reactions?

Why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous storms so people can get their families to safety. Political analysts use patterns to predict election outcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

Model 1 - Types of Reactions

$$N_{y}(g) + 3H_{y}(g) \rightarrow 2NH_{y}(g)$$

$$2SO_j(g) + O_j(g) \rightarrow 2SO_j(g)$$

$$MgO(s) + H_2O(l) \rightarrow Mg(OH)_2(aq)$$

$$\mathrm{P_{2}O_{3}(g)} + 3\mathrm{H_{2}O(l)} \rightarrow 2\mathrm{H_{2}PO_{4}(aq)}$$

$$SO_{*}(g) + H_{*}O(l) \rightarrow H_{*}SO_{*}(aq)$$

serc single replacement reaction

$$2Al(NO_s)_s(aq) + 3Ca(s) \rightarrow 3Ca(NO_s)_s(aq) + 2Al(s)$$

$$2Al(s) + 6HCl(aq) \rightarrow 2AlCl_s(aq) + 3H_s(g)$$

$$Cl_j(g) + 2NaBr(aq) \rightarrow 2NaCl(aq) + Br_j(l)$$

$$ZnBr_{r}(xq) + F_{r}(g) \rightarrow ZnF_{r}(xq) + Br_{r}(l)$$

Son B Decomposition Reaction

$$MgCO_{\epsilon}(s) \rightarrow MgO(s) + CO_{\epsilon}(g)$$

$$8Li_*S(s) \rightarrow 16Li(s) + S_*(s)$$

$$2H_1O(1) \rightarrow 2H_1(g) + O_1(g)$$

$$2KClO_{s}(s) \rightarrow 2KCl(s) + 5O_{s}(g)$$

$$2Na_sO_s(s) \rightarrow 2Na_sO(s) + O_s(g)$$

$$(NH_s)_sCO_s(s) \rightarrow 2NH_s(g) + H_sO(0 + CO_s(g))$$

Ser D Double Replacement Reaction

$$AgNO_s(aq) + NaO((aq) \rightarrow AgO(s) + NaNO_s(aq)$$

$$2HNO_{s}(sq) + Mg(OH)_{s}(sq) \rightarrow Mg(NO_{s})_{s}(sq) + 2H_{s}O(l)$$

$$Na_sCO_s(aq) + CaCL_s(aq) \rightarrow CaCO_s(s) + 2NaCl_s(aq)$$

$$FeS(s) = 2HCl(aq) \rightarrow H_sS(g) + FeCl_s(aq)$$

$$HCl(aq) + NsOH(aq) \rightarrow H_1O(l) + NsCl(aq)$$

The chemical equations in Model 1 contain the phase notations (s), (f), (g), and (aq). Match
each numbel with its meaning.

dissolved in water







- Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)?

 Set

 D
- Based on the examples provided, which set(s) of reactions in Model 1 typically involve gases and/or solids?
 Set. B





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Pier A. de Groot

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treatment of the theory and natural variations of light stable isotopes It discusses isotope distribution in the context of fractionation processes thermodynamics mass conservation exchange kinetics and diffusion theory and includes more than 100 original equations. The theoretical principles are illustrated with natural examples that emphasize oxygen and hydrogen isotope variations in natural waters terrestrial and extraterrestrial rocks and hydrothermal systems. New data on meteoric precipitation rivers springs formation fluids and hydrothermal systems are included in relation to various natural phenomena. Essentially this book seeks to reconnect the diverse phenomenological observations of isotope distribution to the quantitative theories of physical chemistry and the language of differential equations. It may serve as a textbook for advanced students as a research reference or as a quick source of information. The book is organized into five chapters each followed by suggested quantitative problems and a short reference list. The three theoretical chapters progress from an elementary review of the physical chemistry of stable isotopes to the thermodynamics of isotopic compounds and finally to the calculation of isotope distribution in dynamic systems. The third and fifth chapters emphasize oxygen and hydrogen isotope variations in Earth's hydrosphere and lithosphere constituting the most important examples of the theoretical principles Appendices provide data on atomic weights of light elements physical constants mathematical relationships and isotopic fractionation factors

Isotopes in Biology George Wolf, 2013-09-24 Isotope is Biology is a six chapter supplementary text that covers the properties and application of isotopes as labels or analytical tools in biological research. The first chapters deal with the physico chemical properties and radioactivity of isotopes These chapters also explore their synthesis preparation radiation decomposition and decay of radioactivity. The succeeding chapter considers other aspects of isotopes including their effect of health disposal spills and laboratory use Another chapter examines the chemical and biochemical behavior natural abundance and the chemical stability of isotopic compounds The final chapters describe several isotopic methods namely isotope dilution paper chromatography and autoradiography with emphasis on their application in biological studies This book will be of value to biologists and graduate and undergraduate biology students Handbook of Environmental Isotope Geochemistry Mark Baskaran, 2011-10-13 Applications of radioactive and stable isotopes have revolutionized our understanding of the Earth and near earth surface processes The utility of the isotopes are ever increasing and our sole focus is to bring out the applications of these isotopes as tracers and chronometers to a wider audience so that they can be used as powerful tools to solve environmental problems New developments in this field remain mostly in peer reviewed journal articles and hence our goal is to synthesize these findings for easy reference for students faculty regulators in governmental and non governmental agencies and environmental companies While this volume maintains its rigor in terms of its depth of knowledge and quantitative information it contains the breadth needed for wide variety problems and applications in the environmental sciences. This volume presents all of the newer and older applications of isotopes pertaining to the environmental problems in one place that is readily accessible to readers This book not only has the depth and rigor that is

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Radiochemistry and the Discovery of Isotopes Alfred Romer, 1970 Stable Isotope Geochemistry Jochen Hoefs, 2009-01-13 Stable Isotope Geochemistry is an introduction to the use of stable isotopes in the geosciences It is subdivided into three parts theoretical and experimental principles fractionation processes of light and heavy elements the natural variations of geologically important reservoirs Since the application of stable isotopes to earth sciences has grown in the last few years a new edition appears necessary Recent progress in analysing the rare isotopes of certain elements for instance allow the distinction between mass dependent and mass independent fractionations Special emphasis has been given to the growing field of heavy elements Many new references have been added which will enable quick access to recent literature For students and scientists alike the book will be a primary source of information with regard to how and where stable isotopes can be used to solve geological problems <u>Isotope Dilution Analysis</u> J. Tölgyessy, T. Braun, M. Kyrš, 2013-10-22 International Series of Monographs in Analytical Chemistry Volume 49 Isotope Dilution Analysis focuses on the method of isotope dilution analysis IDA The book first discusses the principles types and theory of IDA Classification of the methods of IDA precision accuracy and sensitivity of IDA and types of IDA are described The text also examines experimental techniques separation and mass determination The separation of components reagents and tracers and amount of substance separated are underscored The text takes a look at the inorganic applications of IDA including determination of elements and selected procedures The text examines the applications of IDA in organic chemistry and biochemistry

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expansion of their atomic weight uncertainties For 7 elements magnesium calcium iron zinc molybdenum palladium and tellurium documented isotope abundance variations in materials of natural terrestrial origin are too small to have a Isotopes Rob Ellam, 2016-05-12 An isotope is a variant significant effect on their standard atomic weight uncertainties form of a chemical element containing a different number of neutrons in its nucleus Most elements exist as several isotopes Many are stable while others are radioactive and some may only exist fleetingly before decaying into other elements In this Very Short Introduction Rob Ellam explains how isotopes have proved enormously important across all the sciences and in archaeology Radioactive isotopes may be familiar from their use in nuclear weapons nuclear power and in medicine as well as in carbon dating They have been central to establishing the age of the Earth and the origins of the solar system Combining previous and new research Ellam provides an overview of the nature of stable and radioactive isotopes and considers their wide range of modern applications ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable Handbook of Isotopes in the Cosmos Donald D. Clayton, 2003-09-11 An information resource about the isotopes and their place in the cosmos **Isotopes in the Atomic Age** Hari Jeevan Arnikar,1989

Handbook of Stable Isotope Analytical Techniques Pier A. de Groot,2004-10-27 Parent with price Volume I contains subjective reviews specialized and novel technique descriptions by guest authors Part 1 includes contributions on purely analytical techniques and Part 2 includes matters such as development of mass spectrometers stability of ion sources standards and calibration correction procedures and experimental methods to obtain isotopic fractionation factors Volume II will be available in 2005

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