

NCSD HIGH SCHOOL CHEMISTRY SCOPE AND SEQUENCE

UNIT	Atomic Structure & Properties	Periodic Table	Bonding	Chemical Reactions	Moles / Stoichiometry	Acids and Bases
	15 Days	11 Days	16 Days	16 Days	26 Days	9 Days
S T A N D A R D S	H.1P.1 Explain how atomic structure is related to the properties of elements and their position in the Periodic Table. Explain how the composition of the nucleus is related to isotopes and radioactivity. H.2P.2 Explain how physical and chemical changes demonstrate the law of conservation of mass. H.2P.3 Describe the interactions of energy and matter including the law of conservation of energy.	H.1P.1 Explain how atomic structure is related to the properties of elements and their position in the Periodic Table. Explain how the composition of the nucleus is related to isotopes and radioactivity. H.2P.2 Explain how physical and chemical changes demonstrate the law of conservation of mass.	H.1P.1 Explain how atomic structure is related to the properties of elements and their position in the Periodic Table. Explain how the composition of the nucleus is related to isotopes and radioactivity. H.1P.2 Describe how different types and strengths of bonds affect the physical and chemical properties of compounds.	H.2P.1 Explain how chemical reactions result from the making and breaking of bonds in a process that absorbs or releases energy. Explain how different factors can affect the rate of a chemical reaction. H.2P.2 Explain how physical and chemical changes demonstrate the law of conservation of mass. H.2P.3 Describe the interactions of energy and matter including the law of conservation of energy.	H.2P.2 Explain how physical and chemical changes demonstrate the law of conservation of mass.	H.2P.1 Explain how chemical reactions result from the making and breaking of bonds in a process that absorbs or releases energy. Explain how different factors can affect the rate of a chemical reaction. H.2P.2 Explain how physical and chemical changes demonstrate the law of conservation of mass.

(MHS: Chemistry; RPHS: General Chemistry; CHS: Pre-AP Chemistry)

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AW Rasmussen



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