

(iii) Write the equation for the termination step which produces a hydrocarbon as a product in this reaction.

(1)

(b) Ethene also reacts with chlorine but by a different mechanism.

* (i) Describe how the π bond in ethene forms and explain why this bond causes ethene to take part in addition reactions with halogens.

(2)

* (ii) Write the mechanism for the reaction of ethene with chlorine.

Use curly arrows to show movements of electron pairs.

(3)



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