- 2. What does the number next to isotopes signify? | 1'5 +he mass
- 3. How can you tell isotopes apart? by the number of neutrons

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chromium-58	Chromium-63
# of protons	24	2
# of neutrons		
# of electrons		-

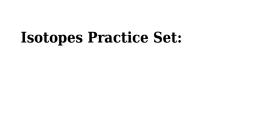
	Carbon-12	Carbon-16
# of protons		
# of neutrons		
# of electrons		_

	Nitrogen-15	Nitrogen-20
# of protons		
# of neutrons		
# of electrons		

	Sulfur-23	Sulfur-25
# of protons		
# of neutrons		
# of electrons		

# **Isotopes Practice Set**

**RD Boyd** 



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