

Ions and Isotopes Worksheet

Symbol	Name of the element	Atomic number	Protons	Neutrons	Electrons	Atomic Mass	Charge
O -17	Oxygen	8	8	9	8	17	0
Xe - 131	Xenon	54	54	77	54	131	0
$^{207}_{82}\text{Pb}$	Lead	82	82	125	82	207	0
Na¹⁺	Sodium ion	11	11	12	10	23	+1
Sr	Strontium	38	38	50	38	88	0
Al³⁺	Aluminum ion	13	13	14	10	27	+3
$^{238}_{92}\text{U}$	Uranium	92	92	146	92	238	0
$^{75}_{33}\text{As}^{-3}$	Arsenic	33	33	39	36	75	-3
$^{32}_{16}\text{S}^{-2}$	Sulfide ion	16	16	16	18	32	-2
Na	Sodium	11	11	12	11	23	0
N³⁻	Nitride ion	7	7	7	10	14	-3
$^{65}_{29}\text{Cu}$	Copper	29	29	36	29	65	0
$^{120}_{80}\text{Hg}$	Mercury isotope	80	80	40	80	120	0
Ag	silver	47	47	61	47	108	0
Zr	zirconium	40	40	51	91	40	0
Cr	chromium	24	24	28	24	56	0
$^{261}_{104}\text{Rf}$	rutherfordium	104	104	158	104	261	0

Isotopes And Ions Worksheet Answer Key

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