SNAP

: Balance the following chemical equations

20.
$$AI + CI_2 \rightarrow AICI_3$$

21. $AI + FeO \rightarrow AI_2O_3 + Fe$
22. $Na_2CO_3 + HCI \rightarrow H_2CO_3 + NaCI$
23. $P_4O_{10} + H_2O \rightarrow H_3PO_4$
24. $AI_2(SO_4)_3 + Ca(OH)_2 \rightarrow AI(OH)_3 + Ca$
25. $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$
26. $Fe_2O_3 + H_2 \rightarrow Fe + H_2O$
27. $N_2O_5 + H_2O \rightarrow HNO_3$
28. $H_3AsO_4 \rightarrow As_2O_5 + H_2O$
29. $Pb + Na + C_3H_3CI \rightarrow Pb(C_3H_5)_4 + Ca$

Part IV: In each of the following problems you are given some potent reaction. Determine what type of reaction would occur, predict the p the complete chemical equation and balance it!

- 30. Aluminum powder burns in the presence of oxygen to form ...
- 31. Magnesium metal reacts with nitric acid to form ...
- 32. Aqueous barium chloride reacts with sodium hydroxide to form ...
- 33. Butane burns in the presence of oxygen to form ...
- 34. Hydrogen gas reacts with chlorine gas to form ...
- 35. Aluminum powder reacts with iron(III) oxide to form...

Formulas And Chemical Reactions Study Guide

JG Myers

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Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing,2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions

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