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web the equation  $\text{mg} + \text{hcl} \rightarrow \text{mgcl}_2 + \text{h}_2$  is clearly unbalanced because on one side there is only 1 hydrogen atom but on the other side there are 2 also unbalanced chlorine but they both come from the same compound so this becomes really simple to balance this equation is easily balanced by placing the coefficient 2 in front of molecule

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 $3 \text{ KClO}_3 \rightarrow 3 \text{ KCl} + 3 \text{ O}_2$   
 $2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$   
 $4 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$   
 $5 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 + 6 \text{ H}_2\text{O}$   
 $3 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 4 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2\text{O}$   
 $4 \text{ CH}_4 + \text{O}_2 \rightarrow 2 \text{ CO}_2 + 2 \text{ H}_2\text{O}$   
 $8 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4$

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