

## Paper 1

Questions are applicable for both core and extended candidates

- 1 Which statement describes methane?
- A It is an alcohol.
  - B It is an unsaturated molecule.
  - C It contains carbon, hydrogen and oxygen atoms only.
  - D Its molecules contain four single covalent bonds.
- 2 Which equation represents the cracking of an alkane?
- A  $3\text{C}_2\text{H}_4 \rightarrow \text{C}_6\text{H}_{12}$
  - B  $\text{C}_6\text{H}_{12} + \text{H}_2 \rightarrow \text{C}_6\text{H}_{14}$
  - C  $\text{C}_6\text{H}_{14} \rightarrow 6\text{C} + 7\text{H}_2$
  - D  $\text{C}_6\text{H}_{14} \rightarrow \text{C}_2\text{H}_4 + \text{C}_4\text{H}_{10}$
- 3 Which statement about ethane is correct?
- A It decolourises bromine water.
  - B It burns in excess oxygen to form water and carbon dioxide.
  - C Its molecular formula is  $\text{C}_2\text{H}_4$ .
  - D Its atoms are joined together by ionic bonding.
- 4 Some properties of colourless liquid L are listed.
- It boils at  $65^\circ\text{C}$ .
  - When added to water, two layers form which do not mix.
  - It does not react with sodium carbonate.
  - It has no effect on bromine water.

What is L?

- A ethanol
- B hexane
- C hexene
- D ethanoic acid

# Igcse Chemistry Examination Paper Jan 24

**Bryan Earl, L. R. Wiford**



## **Igcse Chemistry Examination Paper Jan 24:**

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