

Name: _____ **KEY** _____
Solubility WS

Use your solubility curve chart to answer the following questions.

Part 1: Graph interpretation: (no work needed)

What are the customary units of solubility on solubility curves? **Grams of Solute per 100 g of water**

1. According to the graph, the solubility of any substance changes as **temperature** changes.
2. Most substances on this graph show increased solubility as temperature increases. What are the exceptions? **$\text{Ce}_2(\text{SO}_4)_3$ and NH_3**
3. The solubilities of substances whose curves show greater (steeper) slopes are **(more/less)** affected by temperature changes than those that have more gradual slopes.
4. Which salt has solubility values that are least affected by changes in temperature? **NaCl**
5. What is the solubility of ammonium chloride at 60°C ? **50 g/100 g H_2O**
6. At what temperature do potassium chlorate and potassium chloride have the same solubility in water? **95°C**
7. Which compound is least soluble in water at 12°C ? **KClO_3**
8. A saturated solution of which compound contains 130 grams of solute per 100 grams of water at 70°C ? **KNO_3**
9. Are the following solutions unsaturated, saturated, or supersaturated?
 - a. 80 g of sodium nitrate in 100 g of water at 30°C . **unsaturated**
 - b. 80 g of potassium chlorate in 100 g of water at 50°C . **supersaturated**
10. Which saturated solution of a chloride has the greatest percentage by mass of solute at 60°C ? **don't worry about this question for the test**

Problems: Show all work on a separate sheet.

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AW Rasmussen



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Table of Contents Holt Concentration Of Solutions Page 2answer Key

1. Understanding the eBook Holt Concentration Of Solutions Page 2answer Key
 - The Rise of Digital Reading Holt Concentration Of Solutions Page 2answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Concentration Of Solutions Page 2answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Concentration Of Solutions Page 2answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Concentration Of Solutions Page 2answer Key
 - Personalized Recommendations
 - Holt Concentration Of Solutions Page 2answer Key User Reviews and Ratings

- Holt Concentration Of Solutions Page 2answer Key and Bestseller Lists
- 5. Accessing Holt Concentration Of Solutions Page 2answer Key Free and Paid eBooks
 - Holt Concentration Of Solutions Page 2answer Key Public Domain eBooks
 - Holt Concentration Of Solutions Page 2answer Key eBook Subscription Services
 - Holt Concentration Of Solutions Page 2answer Key Budget-Friendly Options
- 6. Navigating Holt Concentration Of Solutions Page 2answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Concentration Of Solutions Page 2answer Key Compatibility with Devices
 - Holt Concentration Of Solutions Page 2answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Concentration Of Solutions Page 2answer Key
 - Highlighting and Note-Taking Holt Concentration Of Solutions Page 2answer Key
 - Interactive Elements Holt Concentration Of Solutions Page 2answer Key
- 8. Staying Engaged with Holt Concentration Of Solutions Page 2answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Concentration Of Solutions Page 2answer Key
- 9. Balancing eBooks and Physical Books Holt Concentration Of Solutions Page 2answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Concentration Of Solutions Page 2answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Concentration Of Solutions Page 2answer Key
 - Setting Reading Goals Holt Concentration Of Solutions Page 2answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Concentration Of Solutions Page 2answer Key
 - Fact-Checking eBook Content of Holt Concentration Of Solutions Page 2answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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