# CHEMISTRY ADVANCED STUDY

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# FUNDAMENTAL CONCEPTS

#### Math

- · Properties of exponents
- · Properties of logarithms

- a log, (M) = p log, (M)
- · Graphing slope integrals
  - o venue+b
  - b is the y-intercept, m is the slope
- Quadratic formula:

$$0 \quad X \equiv \frac{-b \pm cb^2 - 4ac}{2a}$$

$$0 \quad x = \frac{-4 \pm \sqrt{14 - 4 \pm (-21)}}{2}$$

$$6 x = -2 \pm 5$$

#### Science

- Scientific notation
- Dimensional analysis
- Significant figures

#### Physics and Biology

- Scientific method.
- Coulomb's law: electric force between like or opposite charges; dependence on separation distance and charge magnitude

# ELEMENTS AND ATOMS

Protons: define the element (atomic number), positive

Electrons: define the charge (bonds and polarity), negative

Neutrons: define isotopes, neutral

# ATOMIC MASS AND WEIGHT

Atomic Mass: Carbon-12 (exactly 12 atomic mass unit)

#### Atomic Weight

- · weighted average of atomic masses of element
- (atomic mass neutrons) + (atomic mass neutrons)

# MOLE A.K.A. AVOGADRO'S NUMBER

× 6.02 - 10P

#### In Terms of Carbon

1 mole of PC atoms in 12g of Carbon

6.02 - 10<sup>22</sup> = <sup>12</sup>C atoms in 12g of Carbon

#### Gram to AMU

1 gram = 1 mole of atomic mass units.

1 gram = 6.02 - 10<sup>(2)</sup> amu

1 mai PAI = 27g

2 mail 253 + 56a

4 mol ™O ≈ 64g

# **Holt Chemistry Study Guide Atoms And Moles**

Russell C. Hollingsworth

# **Holt Chemistry Study Guide Atoms And Moles:**

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Library of Congress Catalog:

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S. Camp Russell C. Hollingsworth, 1979 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston
Staff, 2003-03 Space Science, 1969 Scientific, Medical, and Technical Books Published in the United States of
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