Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry Section 1: What is Chemistry? Objectives:

- Describe ways in which chemistry is a part of your daily life.
- Describe the characteristics of three common states of matter.
- Describe physical and chemical changes, and give examples of each.
- Identify the reactants and products in a chemical reaction.
- List four observations that suggest a chemical change has occurred.

A \_\_\_\_\_ is any substance that has a definite composition. A \_\_\_\_\_ is the process by which one or more substances change to produce one or more new substances. The \_\_\_\_\_ of \_\_\_\_ are the physical forms of matter which are solid, liquid, gas, and plasma.

- have a fixed volume and shape that result from the way their particles are arranged.
- have a fixed volume but not a fixed shape.
  have neither fixed volume nor shape.

Water molecule, H<sub>2</sub>O Water molecule, H<sub>2</sub>O Water molecule, H<sub>2</sub>O

Physical changes are changes in which the

of a substance doesn't change.

Changes of state are physical changes.

Chemical changes occur when the \_\_\_\_\_\_ of substances change and new substances form.

Evidence of a chemical change:



without accepta, secol, or vinegast, and soldinam hydrogen carbonate, or baking soda, are mircol, the solution bubbles as carbon dionale forms.



b When solitions of soliton solitate and endinness natrate are mixed, cadmins solitate, a solid proceputate, forms.



 When aluminum nearly with iron(EH) inside in the clay post, emergy is released in heart and light.



d When phenolphthalein is added to announce dissolved in water, a color change from coloritos to pink occurs.

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